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Thiocarbene and alkoxycarbene tungsten complexes exhibit typically different reaction paths $\stackrel{\text{\tiny{}^{\wedge}}}{\to}$

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Abstract

The condensation of (butyl)thiocarbene tungsten complex $[(OC)_5W=C(SEt)Bu]$ (1a) with an α,β -unsaturated *secondary* acid amide R²CH=CHC(=O)NHR¹ 4 in the presence of POCl₃/Et₃N gives cyclopentadienimines 12, whereas the isostructural alkoxycarbene complex $[(OC)_5W=C(OEt)Bu]$ (1c) under similar conditions affords a (*N*-enamino)ethoxycarbene compound 9. Furthermore, condensation of the (methyl)thiocarbene tungsten complex $[(OC)_5W=C(OEt)Me]$ (1b) with an amide 4 yields cyclopentenimines 19 and allenylidene complexes 20, whereas the corresponding ethoxycarbene complex $[(OC)_5W=C(OEt)CH_3]$ (1d) forms 4-*NH*-amino-1-tungsta-1,3,5-hexatrienes 16 under similar conditions.

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1. Introduction

Fischer alkoxycarbene [1] and aminocarbene [2] complexes have found wide application in organic synthesis [3], whereas thiocarbene complexes have gained much less attention [4–7]. The reaction patterns of thiocarbene complexes known so far are different from those of aminocarbene complexes, but closely related to those of alkoxycarbene compounds in so far as, for example, addition of isocyanides to both thiocarbene and alkoxycarbene complexes gives ketenimine complexes and metal-free ketenimines [4b,8a], addition of phosphines affords ylides [8b], addition of aminoalkynes [4d,9] to (aryl)thiocarbene complexes produces (alkenyl)aminocarbene complexes and indenes [9], and addition of terminal alkynes to (aryl)thiocarbene complexes affords naphthalenes [4d]. Alerted by the observation that (alkenyl)thiocarbene complexes [4d] unexpectedly underwent a

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¹ Crystal structure analysis.

base-induced conversion to allenes [10], a transformation that could not be induced with isostructural (alkenyl)ethoxycarbene compounds, our attention was focused more closely to reactivity differences between thiocarbene and alkoxycarbene complexes [11].

2. Results and discussion

Fischer (alkyl)ethoxycarbene are well known to undergo condensation reactions with aldehydes [12a,12b], ketones [13], acid chlorides [12a,14], imines [15], aminoacetales [16], and acid amides [17]. We now report on striking differences in condensation reactions of isostructural thiocarbene and ethoxycarbene complexes with α , β -unsaturated *secondary* acid amides **4**.

2.1. Condensation of thiocarbene complex 1a with α,β -unsaturated secondary acid amides 4a-e

To a mixture of thiocarbene complex 1a and an imidoyl chloride 5a–e, generated *in situ* from an α , β -unsaturated *secondary* acid amides 4a–e and POCl₃/Et₃N at ambient temperature, was added triethylamine at -40 °C to give a

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cyclopentadienimine complex 12a-e, which could be isolated by chromatography on silica gel.

Whereas base-induced condensation reactions of ethoxycarbene complexes 1c,d are readily performed at ambient temperature, thiocarbene complexes 1a,b would rearrange under such conditions to vinyl thioethers 2a,b and 3a,b (Eq. (1)) rather than form a condensation product. In order to obtain reasonable yields of condensation products from a thiocarbene complex and an imidoyl chloride 5 (generated *in situ* from an acid amide 4 with POCl₃/Et₃N in *ca.* 48 h at 20 °C according to ¹H NMR measurements) the reaction must be initiated under carefully controlled conditions by addition of triethylamine at -40 °C.

$$(CO)_{5}W = \underset{1}{C(XEt)}CH_{2}R$$

$$\xrightarrow{NEt_{3}(cat.)} (OC)_{5}W \leftarrow (EtX)CH = CHR$$

$$\xrightarrow{+CO}_{-W(CO)_{6}} (EtX)CH = CHR$$
(1)

1–3	a	b	c	d
X	S	S	0	0
R	<i>n</i> -Pr	Н	<i>n</i> -Pr	Н

Based on the type of reaction products, the thiocarbene complex 1a adds to secondary imidoyl chlorides 5 preferentially in 1,2-fashion to give a β -iminocarbene complex 7 (Scheme 1). Products resulting from 1,4-addition were not detected in this, but were found in related reactions with α , β -unsaturated *tertiary* acid amides [18]. The reactivity patterns of $(\beta$ -imino)thiocarbene complexes 7 are typically different from those of $(\beta$ -imino)ethoxycarbene complexes **6**. Whilst compounds **6** undergo a metalla(di- π -methane) rearrangement to (N-enamino)ethoxycarbene complexes 9, [17d,19] the isostructural compounds 7 quite unexpectedly afforded cyclopentadienimines 12 (Scheme 1). The formation of compounds 12 from complexes 7 is highly regioand stereoselective. According to experimental results and theoretical calculations, alkoxycarbene complexes 6 seem to prefer an associative reaction (involving a nucleo-philic addition to the carbene carbon atom), whereas thiocarbene complexes 7 preferentially undergo a dissociative reaction



^[a] POCl₃/NEt₃, 48 h, 20°C; ^[b] metalla(di-π-methane)skeletal rearrangement; ^[c] π-cyclization;
 ^[d] formed by base-induced rearrangement of compound 12; ^[e] product ratio according to NMR measurements.

Scheme 1. (*N*-Enamino)ethoxycarbene complexes 9 and cyclopentadienimine complexes 12–14 by condensation of ethoxycarbene complex 1c and thiocarbene complex 1a, respectively, with α , β -unsaturated *secondary* acid amides 4.

(by elimination of thiol to give vinylidene compounds 10) [11]. It is suggested that cyclopentadienimines 12 are derived from vinylidene compounds 10 in straight-forward reactions by hydride transfer to give iminium carbonylmetalates 11, which undergo π -cyclization to compounds 12. Based on a series of NMR analyses of product mixtures, it appears that compounds 12 are generated as single isomers, which rearrange during work-up to produce mixtures of compounds 13 and 14 (Eq. (5)).



It should be noted that the intermediate **11** does not collapse to form a pyrrole complex by an α -cyclization (Eq. (6)), as it has been observed in related reactions of thiocarbene complexes with imidoyl chlorides [11], but is sufficiently long-lasting to allow for rotation around the central C–C bond (s. Scheme 1), as requisite for formation of the π -cyclization product **12** in a thermodynamically controlled process.



2.2. Condensation of (methyl)thiocarbene complex 1b with α,β -unsaturated secondary acid amides 4

The condensation of the (methyl)thiocarbene complex **1b**, and (methyl)ethoxycarbene complex **1d**, respectively, with an α , β -unsaturated *secondary* amide **4** takes a course significantly different from that observed with the corresponding (butyl)thiocarbene complex **1a**, and (butyl)ethoxycarbene complex **1c**, respectively (s. Scheme 1), except for the first step, resulting in formation of a (β -imino)thiocarbene complex **15**, and (β -imino)ethoxycarbene complex **17**, respectively (Scheme 2).

Ethoxycarbene complexes 15 form metallahexatrienes 16 which are stabilized by a hydrogen bridge [17d], whereas the thiocarbene complexes 17 lack this type of stabilization and therefore readily undergo π -cyclization to cyclopentenimine complexes 19 (Eq. (9), Scheme 3), and – in competition – an elimination of thiol to yield (deep blue) allenylidene complexes 20 (Eq. (10), Scheme 3) [20]. The conformation required for the π -cyclization of compounds 18 is readily achieved due to the high flexibility of the carbon backbone of this highly polar iminium carbonylmetalate.

2.3. Structure elucidation

The molecular structures of the cyclopentadienes 12, 13 and 14 are based on ¹H and ¹³C NMR data. Compounds 12 can be distinguished from compounds 13 and 14 due to the presence of two olefinic protons. The methylene unit of the cyclopentadienes 13 and 14 exhibits an AB system (13d: $\delta_{\rm H} = 3.07$ and 2.94, ²J = 22.7 Hz; 14d: $\delta_{\rm H} = 3.42$ and 2.76, ²J = 22.7 Hz). The A₁-bands in the IR spectra of compounds 12–14 (e.g. 13a: $\tilde{\nu} = 2068.5$ cm⁻¹) are in a range



^[a] 1,3 H-shift; ^[b] NMR data were not recorded; ^[c] NMR data were recorded from 1:1 mixtures of amides and POCl₃ (yield estimated based on ¹H NMR measurements); ^[d] isolated chemical yield.

Scheme 2. 4-Amino-1-tungsta-1,3,5-hexatrienes 16 and 4-amino-1-tungsta-1,3,5-hexatrienes 18, respectively, by condensation of (methyl)ethoxycarbene complexes 1d and (methyl)thiocarbene complexes 1b, respectively, with α , β -unsaturated *secondary* acid amides 4.



^[a] isolated chemical yields; ^[b] corresponds to the yield of compound 18; ^[c] E/Z-19

equilibration not observed; ^[d] facile decomposition; ^[e] could not be isolated due to hydrolysis on silica gel.

Scheme 3. Competition between the π -cyclization of 4-amino-1-tungsta-1,3,5-hexatrienes 18 to cyclopentenimines 19, and the elimination of ethanethiol to allenylidene complexes 20.

expected for a $W(CO)_5$ unit coordinated to the nitrogen of an imino function [21]. The configurational assignment of the structures 13 and 14 is based on NOE experiments. Irradiation of the methylene protons of the cyclopentadiene 13a results in a positive enhancement of the signal of the $CH_2CH_2CH_3$ protons. The high-field signal of the two $N(CH_3)_2$ proton signals is observed on irradiation of the olefinic CH proton. Compound 14a shows the corresponding NOEs between the CH_2 group and the high-field signal of the two $N(CH_3)_2$ proton signals as well as between the olefinic proton and the CH₂CH₂CH₃ protons. This assignment corresponds to chemical shifts observed for compounds 13c and 14c. The (Z)-configuration is prevailing for steric reasons and the low-field NC H_3 proton signals can be assigned to the (E)-isomers in which the methyl group is directed towards the $W(CO)_5$ unit.

The cyclopentadienimine complex **13a** was characterized also by a crystal structure analysis (Fig. 1). The cyclopentadiene ring of compound **13a** is almost planar [C9–C5– C6–C7 0.5(3)°, C7–C8–C9–C5–0.3(3)°]. The five-membered ring and the plane comprised by the atoms W, N and C2 are strongly distorted against each other [W–N1–C5–C6– 94.2(3)°, W–N1–C5–C9–82.8(3)°]. The C=N bond distance of 1.289(4) Å in compound **13a** is shorter than in the iminium carbonylmetalate **18b** [C5–N1 1.328(3) Å]. Compounds **18, 19** and **20** are easily distinguished by the position of the A₁-bands in the IR spectra: compounds **18** (e.g. **18f**: $\tilde{\nu} = 2060.9 \text{ cm}^{-1}$), characteristical of 1-tungsta-1,3,5-hexatrienes, compounds **19** (e.g. (Z)-**19g**: $\tilde{\nu} = 2065.8 \text{ cm}^{-1}$),



Fig. 1. Molecular structure of the cyclopentadienimine complex **13a** with selected bond lengths (Å), bond angles (°) and dihedral angles (deg): W– N1 2.286(2), N1–C2 1.289(4), N1–C5 1.446(4), C5–C6 1.340(4), C6–C7 1.511(4), C7–C8 1.495(4), C8–C9 1.361(4), C9–C5 1.460(4), C8–C81 1.465(4), C5–N1–C2 116.9(3), C5–N1–W 111.0(2), W–N1–C2 132.2(2), N1–C2–C3 123.4(3), N1–C2–C4 121.2(3), C3–C2–C4 115.3(3), N1–C5–C9 122.2(3), N1–C5–C6 126.3(3), C6–C5–C9 111.5(3), C5–C6–C7 107.0(3), C6–C7–C8 104.8(3), C7–C8–C9 108.3(3), C8–C9–C5 108.5 (3), W–N1–C5–C6 94.2(3), W–N1–C5–C9 –82.8(3), C5–N1–C2–C4 178.4(3), C5–N1–C2–C3 –2.9(4), C9–C5–C6–C7 0.5(3), C7–C8–C9–C5 –0.3(3), C7–C8–C81–C82 –170.0(3).

characteristical of σ -coordination to C=NR group. The very small A₁-bands of the allenylidene complexes **20f**-e are shifted to exceptionally high wave numbers in a very nar-



Fig. 2. Molecular structure of the 4-amino-1-tungsta-1,3,5-hexatriene **18b** with selected bond lengths (Å), bond angles (°) and dihedral angles (°): W–C1 2.280(3), C1–C4 1.390(4), C4–C5 1.417(4), C5–C6 1.469(4), C6–C7 1.332(4), C7–C8 1.466(4), C5–N1 1.328(3), N1–C14 1.473(3), C1–S1 1.742(3), W–C1–S 106.7(1), S–C1–C4 116.2(2), W–C1–C4 137.0(2), C1–C4–C5 129.0(3), C4–C5–N1 122.8(2), C6–C5–N1 117.7(2), C4–C5–C6 119.5(2), C5–C6–C7 123.9 (3), C6–C7–C8 127.1(3), C5–N1–C14 128.6(2), W–C1–C4–C5 3.6(4), S–C1–C4–C5 179.2(2), C1–C4–C5–N1 0.1(4), C1–C4–C5–C6 –178.5(2), C4–C5–C6–C7 –26.4(4), C5–C6–C7–C8 179.3(3), C6–C7–C8–C9 180.0(3), C4–C5–N1–C14 175.0(2), C5–N1–C14–C15 162.5(3).

row range of 2082.8 to 2083.0 cm⁻¹ (e.g. **20f**: $\tilde{v} = 2083.0$ cm⁻¹) [22]. The configuration of compounds (*Z*)-**19** and (*E*)-**19** was assigned by NOE experiments of compounds **19f** and **19j**. While the (*Z*)-isomers showed NOE between *N*-CH₃ and the CH₂ group of the cyclopentene ring, the expected NOE of the (*E*)-isomer between N-CH₃ and the olefinic CH proton could also be observed (see Fig. 2).

The ligand backbone W–C1–C4–C5–N1 of compound **18b** is essentially planar. Its bond angles W–C1–C4 (137.0(2)°), C1–C4–C5 (129.0(3)°), C4–C5–N1 (122.8(2)°) and C5–N1–C14 (128.6(2)°) are above 120°. The N*H* proton is located between two neighbouring carbonyl carbon atoms. This 4-amino-1,3,5-metallatriene shows a carbiminium carbonylmetalate unit, which is characterized by a pattern of alternating bond distances of the W–C=C–C=N⁺ backbone [W–C1 2.280(3) Å, C1–C4 1.390(4) Å, C4–C5 1.417(4) Å, C5–N1 1.328(3) Å]. The W–C1–C4–C5 portion [dihedral angle 3.6(4)°] and the C1–C4–C5–N1 portion [dihedral angle 0.1(4)°] of the molecule both adopt a *cis* configuration. The C4–C5–C6–C7 unit [dihedral angle $-26.4(4)^{\circ}$] exhibits a *cisoidal* arrangement.

3. Conclusion

The reactivity patterns of alkylcarbene complexes 1a-dwith α,β -unsaturated *secondary* acid amides R²CH= CHC(=O)NHR¹ 4 in the presence of POCl₃/Et₃N have been unravelled, and it has been shown that β -iminocarbene complexes are key-intermediates in these reactions. The reactivity of a (β -imino)*thio*carbene complexes is significally different from that of an isostructural (β-imino)ethoxycarbene complex. Furthermore, fundamental changes in reactivities are observed also within the groups of (Bimino)heterocarbene complexes derived from methylcarbene, and alkylcarbene complexes other than methylcarbene complexes, respectively. The condensation of (n-butyl)thiocarbene complex $[(OC)_5W=C(SEt)(n-Bu)]$ (1a) with α,β -unsaturated secondary acid amides 4 yields cyclopentadienimine complexes 12-14 (reaction initiated by 1,2addition and subsequent π -cyclization), whereas the (methyl)thiocarbene complex $[(OC)_5W=C(SEt)CH_3]$ (1b) under similar conditions affords cyclopentenimine complexes 19 and allenvlidene complexes 20 (reaction initiated by 1,2-addition and aubsequent hydrogen shift). The formation of different products from β -iminocarbene complexes is attributed to minor differences in activation energies of the single steps involved in the multistep-reaction sequences. In line with theoretical predictions [11], the observation of allenvlidene complexes 20 indicates that a dissociative reaction step is favoured with thiocarbene complexes over an associative process.

4. Experimental

All operations were carried out under an atmosphere of argon. All solvents were dried and distilled prior to use. All ¹H and ¹³C NMR spectra were routinely recorded in $CDCl_3$ or C_6D_6 on a Bruker ARX 300 instrument. COSY, HMQC, HMBC, TOCSY and NOE experiments were performed on either a Bruker AMX 400, a Varian 500 Inova or a Varian 600 unity plus instrument. The chemical shifts are given in ppm with TMS ($\delta = 0$ ppm) and the residue signal of CDCl₃ and C₆D₆ ($\delta = 77$ and 128 ppm) as the internal standards for ¹H and ¹³C NMR spectra. IR spectra were measured on a Bruker Vector 22 FT-IR spectrometer. EI and ESI mass spectra were obtained on a doublefocussing Sektorfeld-MS MAT8200 (Thermo-Finnigan-MAT, Bremen) and QUATTRO LCZ (Waters-Micromass, Manchester, UK) spectrometers. HRMS spectra were determined on a MicroTof (Bruker Daltronics, Bremen) instrument with loop injection; for mass calibration sodium formate clusters were used. Elemental analyses were determined on an elementar vario EL III instrument. Analytical TLC plates, Merck TLC aluminium sheets Silica gel 60 F_{254} , were viewed by UV light (254 nm) and stained by iodine. R_f values refer to TLC tests. Merck Silica gel 60 F was used for column chromatography. Flash chromatography was performed under an argon atmosphere. Compound **1a** was prepared according to the literature [11].

4.1. Pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten (1b)

(The procedure given is an adjustment of the procedure presented by Aumann and Schröder [4b] to the special requirements of (alkyl)thiocarbene complexes of tungsten; see Eq. (1)). To pentacarbonyl[1-(ethoxy)eth-1-ylidene]- tungsten (1d, 7.92 g, 20.0 mmol) and sodiumcarbonate (2.12 g, 20.0 mmol) in 150 mL dry methanol in a 250-mL round bottom flask is added ethanethiol (1.4 mL, 22.1 mmol) at -40 °C. The progress of the reaction can be monitored by TLC or by IR measurements. On addition of phosphoric acid (2.16 g, 22.0 mmol) after 3 h stirring at -40 °C the color of the solution turns from yellow to red. After addition of 75 mL of water the red carbene complex is extracted with *n*-pentane. Crystallization on dry ice yields analytically pure compound 1b (7.00 g, 85%, $R_{\rm f} = 0.3$ in *n*-pentane) in a 10:1 Z/E-ratio.



4.1.1. Data for (Z)-1b {(E)-1b, ca. 10%}

¹H NMR (300 MHz, CDCl₃, 298 K): δ 3.37 {3.43} (s, 3H; CCH₃), 3.02 {3.64} (q, ³*J*(H,H) = 7.5 Hz {7.5 Hz}, 2H; SC*H*₂CH₃), 1.37 {1.54} (t, ³*J*(H,H) = 7.5 Hz {7.5 Hz}, 3H; SCH₂CH₃); ¹³C NMR (75 MHz, CDCl₃, 298 K): δ 332.2 {[23]} (C_q; W=C), 207.3 and 197.6 {[23]} [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 47.7 {51.8} (CH₃; CCH₃), 37.2 {42.4} (CH₂; SCH₂CH₃), 11.5 {12.4} (CH₃; SCH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{\nu} = 2066.4$ (40) {2058.9 (4)}, 2058.9 (2), 1950.0 (100) [ν (C=O)]; MS (70 eV): *m*/*z* for ¹⁸⁴W (%): 412 (20) [*M*]⁺, 486 (10) [*M* - CO]⁺, 328 (35), 298 (35), 270 (100), 127 (50); elemental analysis (%) calcd for C₉H₈O₅SW (412.1): C, 26.23; H, 1.96; found: C, 26.05; H, 1.70%.

4.2. Pentacarbonyl[1-(ethylsulfanyl)pent-1-ene, S-W]tungsten (2a), 1-(ethylsulfanyl)pent-1-ene (3a)

To pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (1a, 113 mg, 0.25 mmol) in 1 mL of CH₂Cl₂ is added triethylamine (10 mg, 0.1 mmol). Within 15 min at 20 °C the solution turned from dark red to yellow. Solvent was removed to give a 5:1 mixture of isomeric compounds (Z)-2a and (E)-2a [R_f (1Z)-2a = 0.5, R_f (1E)-2a = 0.4 in *n*pentane, yellow oil], which are spontaneously transformed into compounds 3a by addition of one equivalent of pyridine.

4.2.1. Data for $(Z)-2a\{(E)-2a\}$

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.97 {6.10} (dt, ³*J*(H,H) = 8.8 Hz {14.7 Hz}, ³*J*(H,H) = 7.4 Hz {7.1 Hz}, 1H; 2-H), 5.88 {5.86} (dt, ³*J*(H,H) = 8.8 Hz {14.7 Hz}, ⁴*J*(H,H) = 1.3 Hz {1.4 Hz}, 1H; 1-H), 2.84 {2.84} (q, ³*J*(H,H) = 7.4 Hz {7.4 Hz}, 2H; SCH₂), 2.30 {2.21} (m, 2H; 3-H₂), 1.47 {1.47} (m, 2H; 4-H₂), 1.28 {1.29} (t, ${}^{3}J(\text{H},\text{H}) = 7.4 \text{ Hz}$ {7.4 Hz}, 3H; SCH₂CH₃), 0.96 {0.94} (t, ${}^{3}J(\text{H},\text{H}) = 7.1 \text{ Hz}$ {7.4 Hz}, 3H; 5-H₃); ${}^{13}\text{C}$ NMR (100 MHz, CDCl₃, 300 K): δ 200.2 and 197.1 {200.4 and 197.2} [each C_q, 1:4, *trans-* and *cis-*CO; W(CO)₅], 140.2 {142.6} (CH; C2), 124.3 {122.9} (CH; C1), 40.1 {39.6} (CH₂; SCH₂), 31.0 {34.3} (CH₂; C3), 22.0 {21.9} (CH₂; C4), 14.8 {14.3} (SCH₂CH₃), 13.6 {13.5} (CH₃; C5); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{\nu} = 2074.0$ (10), 1939.9 (100), 1928.7 (40) [ν (C=O)]; IR (diffuse reflexion) [cm⁻¹ (%)]: $\tilde{\nu} = 2073.3$ (10), 1890.0 (100) [ν (C=O)], 1454.1 (5); MS (70 eV): *m/z* for ¹⁸⁴W (%): 454.0 (30) [*M*]⁺, 426.0 (10) [*M* - CO]⁺, 370.0 (10) [*M* - 3CO]⁺, 341.0 (100); HRMS (ESI) calcd for C₁₂H₁₄SO₅WNa [*M* + Na]⁺: 476.9963; found: 476.9948.

4.2.2. Data for (Z)-3a {(E)-3a}

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.93 {5.92} (dt, ³*J*(H,H) = 9.4 Hz {15.1 Hz}, ⁴*J*(H,H) = 1.4 Hz {1.4 Hz}, 1H; 1-H), 5.57 {5.63} (dt, ³*J*(H,H) = 9.4 Hz {15.1 Hz}, ³*J*(H,H) = 7.2 Hz {7.0 Hz}, 1H; 2-H), 2.66 {2.65} (q, ³*J*(H,H) = 7.4 Hz {7.4 Hz}, 2H; SCH₂), 2.09 {2.06} (m, 2H; 3-H₂), 1.41 {1.41} (m, 2H; 4-H₂), 1.28 {1.27} (t, ³*J*(H,H) = 7.4 Hz {7.4 Hz}, 3H; SCH₂CH₃), 0.91 {0.91} (t, ³*J*(H,H) = 7.4 Hz {7.4 Hz}, 3H; 5-H₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 129.4 {130.8} (CH; C2), 124.5 {122.4} (CH; C1), 31.0 {35.1} (CH₂; C3), 27.6 {26.6} (CH₂; SCH₂), 22.0 {22.3} (CH₂; C4), 15.3 {14.4} (SCH₂CH₃), 13.5 {13.3} (CH₃; C5).

4.3. Pentacarbonyl[(ethylsulfanyl)ethene, S-W]tungsten (0) (2b), (ethylsulfanyl)ethene (3b)

To pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 103 mg, 0.25 mmol) in 1 mL of CH₂Cl₂ is added triethylamine (10 mg, 0.1 mmol). After 4 h, 20 °C the solution turns from red to yellow. Evaporation of solvent gives compound **2b** ($R_f = 0.3$ in *n*-pentane, yellow oil) as the only product, which is transformed into compound **3b** by addition of one equivalent of pyridine.

4.3.1. Data for 2b

¹H NMR (400 MHz, C₆D₆, 300 K): δ 5.47 (dd, ³*J*(H,H) = 16.3 Hz, ³*J*(H,H) = 9.3 Hz, 1H; 1-H), 4.91 (d, ²*J*(H,H) = 0.8 Hz, ³*J*(H,H) = 16.3 Hz, 1 H; *cis*-2-H), 4.79 (d, ²*J*(H,H) = 0.8 Hz, ³*J*(H,H) = 9.3 Hz, 1H; *trans*-2-H), 2.06 (q, ³*J*(H,H) = 7.3 Hz, 2H; SCH₂), 0.62 (t, ³*J*(H,H) = 7.3 Hz, 3H; SCH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.0 and 197.4 [each C_q, 1:4, *trans*and *cis*-CO; W(CO)₅], 132.0 (CH; C1), 121.9 (CH; C2), 38.2 (CH₂; SCH₂), 13.9 (SCH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{\nu}$ = 2075.1 (10), 1941.9 (100), 1930.9 (40) [ν (C==O)]; MS-ESI (ESI): *m*/*z* (%) = 411.1 (100) [*M* – H]⁻.

4.3.2. Data for **3b**

¹H NMR (400 MHz, C₆D₆, 300 K): δ 6.17 (dd, ³*J*(H,H) = 16.7 Hz, ³*J*(H,H) = 10.2 Hz, 1H; 1-H), 5.02 (d, ³*J*(H,H) = 10.2 Hz, 1H; *cis*-2-H), 4.98 (d, ³*J*(H,H) = 16.7 Hz, 1H; *trans*-2-H), 2.31 (q, ³*J*(H,H) = 7.4 Hz, 2H; SCH₂), 0.99 (t, ³*J*(H,H) = 7.4 Hz, 3H; SCH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 132.9 (CH; C1), 110.1 (CH; C2), 25.3 (CH₂; SCH₂), 14.1 (SCH₂CH₃).

4.4. Pentacarbonyl[isopropylidene-(4-phenyl-2-propylcyclopenta-1,4-dienyl)-amine, N-W]tungsten(0) (13a), pentacarbonyl[isopropylidene-(4-phenyl-2-propylcyclopenta-1,3-dienyl)- amine, N-W]tungsten(0) (14a)

(E)-N-Isopropyl-3-phenyl-acrylimidoyl chloride (5a) (generated in situ by addition of (2E)-N-isopropyl-3-phenyl acrylamide (4a, 378 mg, 2.0 mmol) to phosphorous oxychloride (306 mg, 2.0 mmol) in dry dichloromethane (2 mL) in a 3-mL screw-top vessel at 20 °C, 48 h) was added to pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (1a, 454 mg, 1.0 mmol) in dry dichloromethane (3 mL). A color change to dark blue is observed while triethylamine (404 mg, 4.0 mmol) is added at -40 °C. On warming to 20 °C after 10 min, the color fades to yellow. Evaporation of the solvent and flash chromatography at 20 °C on silica gel (40×1 cm, 10:1 *n*-pentane/diethyl ether) affords a bright yellow 4:1 mixture of compounds 13a (260 mg, 46%, $R_{\rm f} = 0.5$ in 10:1 *n*-pentane/diethyl ether) and 14a (65 mg, 12%, $R_{\rm f} = 0.4$ in 10:1 *n*-pentane/diethyl ether). A more polar colorless fraction contained compound 21a (105 mg, 45%, $R_{\rm f} = 0.3$ in *n*-pentane/diethyl ether 10:1).



4.4.1. Spectroscopic data of 13a (obtained from a 4:1 mixture of compounds 13a and 14a)

¹H NMR (300 MHz, CDCl₃, 303 K): δ 7.48 (m, 2H; *o*-CH Ph), 7.33 (m, 2 H; *m*-CH Ph), 7.22 (m, 1H; *p*-CH Ph), 6.59 (s, 1H; 5-H, NOE (+) with 6'-NCH₃), 3.59 and 3.52 (AB system, ²*J*(H,H) = 23.0 Hz, 2H; 3-H₂, NOE (+) with CH₂CH₂CH₃), 2.52 (s, 3H; 6-H₃), 2.18 (m, 2H; CH₂CH₂CH₃), 2.02 (s, 3H; 6'-H₃), 1.60 (m, 2H; CH₂CH₂CH₃), 0.99 (t, 3H; CH₂CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃, 303 K): δ 202.6 and 198.5 [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 184.0 (C_q; C=N), 152.3 (C_q; C1), 144.9 (C_q; C4), 135.4 (C_q; *i*-C Ph), 128.8 (CH; *m*-C Ph), 127.3 (CH; *p*-C Ph), 126.5 (C_q; C2), 124.9 (CH; C5), 124.8 (CH; *o*-C Ph), 40.5 (CH₂; C3), 31.9 (CH₃; C6), 29.7 (CH₂CH₂CH₃), 23.7 (CH₃; C6'), 22.0 (CH₂CH₂CH₃), 14.7 (CH₂CH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]; $\tilde{\nu} = 2068.5$

(10), 1969.4 (5), 1927.9 (100), 1913.1 (40) [ν (C=O)]; HRMS (ESI) calcd for C₂₂H₂₁NO₅WNa [M + Na]⁺: 586.0825; found: 586.0859; HRMS (ESI) calcd for C₂₀H₂₀NO₃W [M – 2CO – H]⁻: 506.0961; found: 506.0996; elemental analysis (%) calcd for C₂₂H₂₁NO₅W (563.1): C, 46.91; H, 3.76; N, 2.49; found: C, 46.98; H, 3.81; N, 2.36%.

4.4.2. Molecular structure analysis of bf 13a (code 3339.AUM)

Formula $C_{22}H_{21}NO_5W$, $M_r = 563.25 \text{ g mol}^{-1}$, yellow crystal, $0.25 \times 0.20 \times 0.10 \text{ mm}$, a = 10.061(1), b = 10.615(1), c = 11.743(1) Å, $\alpha = 71.60(1)^\circ$, $\beta = 82.21(1)^\circ$, $\gamma = 68.48(1)^\circ$, V = 1106.8(2) Å³, $\rho_{\text{calcd}} = 1.690 \text{ g cm}^{-3}$, $\mu = 52.49 \text{ cm}^{-1}$, empirical absorption correction ($0.354 \le T \le 0.622$), Z = 2, triclinic, space group $P\overline{1}$ (no. 2), $\lambda = 0.71073$ Å, T = 198 K, ω and φ scans, 7275 reflections collected ($\pm h, \pm k, \pm l$), $[(\sin\theta)/\lambda]_{\text{max}} = 0.66 \text{ Å}^{-1}$, 5219 independent ($R_{\text{int}} = 0.021$) and 4869 observed reflections [$I \ge 2\sigma(I)$], 265 refined parameters, R = 0.023, $wR_2 = 0.057$, max./min. residual electron density 0.71/-1.41 e Å^{-3}, hydrogen atoms calculated and refined as riding atoms [24].

4.4.3. Data for 14a

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.44 (m, 2H; *o*-CH Ph), 7.33 (m, 2H; *m*-CH Ph), 7.20 (m, 1H; *p*-CH Ph), 6.73 (s, 1H; 3-H, NOE (+) with $CH_2CH_2CH_3$), 3.89 and 3.28 (AB system, ²*J*(H,H) = 22.7 Hz, 2H; 5-H₂, NOE (+) with 6'-NCH₃ on irradiation at 3.28), 2.52 (s, 3H; 6-H₃), 2.10 (m, 2H; $CH_2CH_2CH_3$), 2.02 (s, 3H; 6'-H₃), 1.60 (m, 2 H; $CH_2CH_2CH_3$), 0.99 (t, 3H; $CH_2CH_2CH_3$); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.4 and 198.4 [each C_q , 1:4, *trans*- and *cis*-CO; W(CO)₅], 184.8 (C_q ; C=N), 151.8 (C_q ; C1), 139.4 (C_q ; C4), 135.1 (C_q ; *i*-C Ph), 130.1 (C_q ; C2), 128.7 (CH; *m*-CH Ph), 127.3 (CH; C3), 127.0 (CH; *p*-CH Ph), 124.6 (CH; *o*-CH Ph), 41.1 (CH₂; C5), 32.0 (CH₃; C6), 28.8 ($CH_2CH_2CH_3$), 23.9 (CH₃; C6'), 20.9 ($CH_2CH_2CH_3$), 14.5 ($CH_2CH_2CH_3$).

4.5. Pentacarbonyl[cyclohexylidene-(4-phenyl-2-propylcyclopenta-1,4-dienyl)-amine, N-W]tungsten(0) (13b), pentacarbonyl[cyclohexylidene-(4-phenyl-2-propylcyclopenta-1,3-dienyl)- amine, N-W]tungsten(0) (14b)

(*E*)-*N*-Cyclohexyl-3-phenyl-acrylimidoyl chloride (**5b**), generated *in situ* from (2*E*)-*N*-cyclohexyl-3-phenyl acrylamide (**4b**, 458 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) in dry dichloromethane (2 mL) at 20 °C, 48 h in a 3-mL screw-top vessel, was added to pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (**1a**, 454 mg, 1.0 mmol) in dry dichloromethane (3 mL). Triethylamine (404 mg, 4.0 mmol) was added at -40 °C and the mixture turned dark blue. The color faded to yellow on warming to 20 °C after 10 min. Evaporation of the solvent and flash chromatography at 20 °C on silica gel (40 × 1 cm, 10:1 *n*-pentane/diethyl ether) afforded a bright yellow 10:1 mixture of compounds **13b** (183 mg, 30%, $R_{\rm f} = 0.6$ in 10:1 *n*-pentane/diethyl ether) and 14b (18 mg, 3%, $R_{\rm f} = 0.5$ in 10:1 *n*-pentane/diethyl ether).



4.5.1. Data for 13b

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.48 (m, 2H; o-CH Ph), 7.33 (m, 2H; *m*-CH Ph), 7.22 (m, 1H; *p*-CH Ph), 6.59 (s, 1H; 5-H), 3.55 (s, 2H; 3-H₂), 2.89 (m, 2H; 6-H₂), 2.37 (m, 2H; 6'-H₂), 1.93 and 1.67 (each m, 2:4 H; 7-, 7'and 8-CH₂), 2.18 (m, 2H; CH₂CH₂CH₃), 1.55 (m, 2H; $CH_2CH_2CH_3$), 0.99 (t, 3H; $CH_2CH_2CH_3$); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.8 and 198.5 [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 189.6 (C_a; C=N), 151.4 (Cq; C1), 144.6 (Cq; C4), 135.4 (Cq; i-C Ph), 128.7 (CH; *m*-C Ph), 127.2 (CH; *p*-C Ph), 126.6 (C_a; C2), 125.5 (CH; C5), 124.8 (CH; o-C Ph), 42.1 (CH₂; C6), 40.5 (CH₂; C3), 33.0 (CH₂; C6'), 29.7 (CH₂CH₂CH₃), 27.9 and 27.7 (CH₂; C7 and C7'), 25.1 (CH₂; C8), 22.2 (CH₂CH₂CH₃), 14.6 (CH₂CH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{v} =$ 2068.0 (10), 1968.5 (3), 1939.7 (25), 1928.2 (100), 1911.6 (40) [v(C=0)]; MS (70 eV): m/z for ¹⁸⁴W (%): 603.3 (2) $[M]^+$, 547 (5) $[M - 2CO]^+$, 279 (50) $[M - W(CO)_5]^+$, 250 (100) $[M - W(CO)_5 - C_2H_5]^+$; HRMS (ESI) calcd for C₂₃H₂₄NO₃W $[M - 2CO - H]^{-}$: 546.1264; found: 546.1292; elemental analysis (%) calcd for C₂₅H₂₅NO₅W (544.3): C 49.77, H 4.18, N 2.32; found C 49.83, H 4.22, N 2.17.

4.5.2. Data for 14b

¹H NMR (300 MHz, CDCl₃, 303 K; for phenyl and cyclohexyl signals see **13b**): δ 6.73 (s, 1H; 3-H), 4.01 and 3.18 (AB system, ²*J*(H,H) = 22.6 Hz, 2H; 5-H₂), 0.88 (t, 3H; CH₂CH₂CH₃); ¹³C NMR (75 MHz, CDCl₃, 303 K): δ 202.6 and 198.4 [each C_q, 1:4, *trans-* and *cis-*CO; W(CO)₅], 190.6 (C_q; C=N), 150.9 (C_q; C1), 139.2 (C_q; C4), 135.2 (C_q; *i-*C Ph), 130.2 (C_q; C2), 128.6 (CH; *m*-CH Ph), 127.4 (CH; C3), 126.9 (CH; *p*-CH Ph), 124.5 (CH; *o-*CH Ph), 42.3 (CH₂; C6), 41.7 (CH₂; C5), 33.3 (CH₂; C6'), 28.8 (CH₂CH₂CH₃), 27.9 and 27.7 (CH₂; C7 and C7'), 25.8 (CH₂; C8), 21.2 (CH₂CH₂CH₃), 14.5 (CH₂CH₂CH₃); for IR and MS data see compound **13b**.

4.6. Pentacarbonyl[(1-methyl-propylidene)-(4-phenyl-2-propyl-cyclopenta-1,4- dienyl)-amine, N-W]tungsten(0) (13c), pentacarbonyl[(1-methyl-propylidene)-(4-phenyl-2-propyl-cyclopenta-1,3- dienyl)-amine, N-W]tungsten(0) (14c)

(*E*)-*N*-sec-Butyl-3-phenyl-acrylimidoyl chloride (5c), generated from (2E)-*N*-sec-butyl-3-phenyl acrylamide (4c, 378 mg, 2.0 mmol) with phosphorous oxychloride

(306 mg, 2.0 mmol) in dry dichloromethane (2 mL) at 20 °C, 48 h in a 3-mL screw-top vessel, was added to pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (1a, 454 mg, 1.0 mmol) in dry dichloromethane (3 mL). Triethylamine (404 mg, 4.0 mmol) was added to the mixture at -40 °C leading to a color change from red to dark blue. On warming after 10 min to 20 °C, the color faded to yellow. Evaporation of the solvent and flash chromatography at 20 °C on silica gel (40 × 1 cm, 10:1 *n*-pentane/diethyl ether) afforded a bright yellow fraction of a 12:2:6:3 mixture of compounds (*Z*)-13c, (*E*)-13c, (*Z*)-14c, (*E*)-14c (370 mg, 64%, $R_f = 0.4$ in 50:1 *n*-pentane/diethyl ether).



Data for (E/Z)-13c and (E/Z)-14c (obtained from a 13:2:6:3 mixture of compounds (Z)-13c, (E)-13c, (Z)-14c, (E)-14c; only some characteristic signals are given for minor isomers).

4.6.1. Data for (Z)-13c

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.48 (m, 2H; o-CH Ph), 7.31 (m, 2 H; m-CH Ph), 7.20 (m, 1H; p-CH Ph), 6.57 (s, 1H; 5-H), 3.57 and 3.50 (AB system, ${}^{2}J(H,H) = 22.8$ Hz, 2H; 3-H₂), 2.78 (m, 2H; 6-H₂), 2.16 (m, 2H; CH₂CH₂CH₃), 1.95 (s, 3H; 6'-H₃), 1.60 (m, 2H; CH₂CH₂CH₃), 1.28 (t, ${}^{3}J$ (H,H) = 7.5 Hz, 3H; 7-H₃), 0.99 (t, ${}^{3}J(H,H) = 7.4$ Hz, 3H; CH₂CH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.5 and 198.4 [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 188.3 (C_q; C=N), 152.4 (C_q; C1), 144.9 (C_q; C4), 135.3 (C_q; *i*-C Ph), 128.6 (CH; *m*-C Ph), 127.1 (CH; *p*-C Ph), 126.3 (C_a; C2), 124.8 (CH; C5), 124.7 (CH; o-C Ph), 40.4 (CH₂; C3), 38.1 (CH₂; C6), 29.5 (CH₂CH₂CH₃), 22.0 (CH₂CH₂CH₃), 20.3 (CH3; C6'), 14.6 (CH2CH2CH3), 11.5 (CH3; C7); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2068.4 (10), 1969.4 (5), 1927.0$ (100), 1912.4 (40) $[v(C \equiv O)]$; HRMS (ESI) calcd for $C_{21}H_{22}NO_3W[M - 2CO - H]^-$: 520.1107; found: 520.1124.

4.6.2. Data for (E)-13c

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.56 (s, 1H; 5-H), 2.45 (s, 3H; 6-H₃).

4.6.3. Data for (Z)-14c

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.72 (s, 1H; 3-H), 3.96 and 3.27 (AB system, ²*J*(H,H) = 22.6 Hz, 2H; 5-H₂), 1.93 (s, 3H; 6'-H₃).

4.6.4. Data for (E)-14c

¹H NMR (400 MHz, CDCl₃, 300 K): δ 3.98 and 3.22 (AB system, ²*J*(H,H) = 22.6 Hz, 2H; 5-H₂), 2.44 (s, 3H; 6-H₃).

4.7. Pentacarbonyl[isopropylidene-(3-methyl-5-propylcyclopenta-1,4-dienyl)- amine, N-W]tungsten(0) (12d), pentacarbonyl[isopropylidene-(4-methyl-2-propylcyclopenta-1,4-dienyl)- amine, N-W]tungsten(0) (13d), pentacarbonyl[isopropylidene-(4-methyl-2-propylcyclopenta-1,3-dienyl)- amine, N-W]tungsten(0) (14d)

(E)-N-Isopropyl-but-2-ene-1-carboximidoyl chloride (5d), generated from (E)-but-2-enonic acid isopropylamide (4d, 254 mg, 2.0 mmol) with phosphorous oxychloride (306 mg, 2.0 mmol) in dry dichloromethane (2 mL) at 20 °C, 48 h in a 3-mL screw-top vessel, was added to pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (1a, 454 mg, 1.0 mmol) in dry dichloromethane (3 mL). At -40 °C triethylamine (404 mg, 4.0 mmol) was added leading to a color change from red to dark blue. The color faded to yellow on warming to 20 °C after 10 min. Evaporation of the solvent and flash chromatography at 20 °C on silica gel $(40 \times 1 \text{ cm}, 10:1 \text{ n-pentane/diethyl ether})$ afforded a bright yellow 6:10:1 mixture of compounds 12d, 13d and 14d (160 mg, 32%, $R_f = 0.6$ in 10:1 *n*-pentane/diethyl ether). Compound 12d in CDCl₃ is transformed completely into a 10:1 mixture of compounds 13d and 14d in 24 h, 20 °C.



Data for 12d, 13d and 14d (obtained from a 6:10:1 mixture of compounds 12d, 13d and 14d).

4.7.1. Data for 12d

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.15 (m, 1H; 4-H), 5.60 (m, 1H; 2-H), 3.21 (m, 1H; CHCH₃), 2.50 (s, 3H; 6-H₃), 1.94 (s, 3H; 6'-H₃), 2.12 and 1.81 (each m, each 1H; CH₂CH₂CH₃), 1.62 (m, 2H; CH₂CH₂CH₃), 1.19 (d, ³*J*(H,H) = 7.7 Hz, 3H; CHCH₃), 0.99 (t, ³*J*(H,H) = 7.4 Hz, 3H; CH₂CH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.0 and 198.7 [each C_q, 1:4, *trans-* and *cis-*CO; W(CO)₅], 183.8 (C_q; C=N), 156.9 (C_q; C1), 140.8 (C_q; C5), 136.1 (CH; C4), 121.2 (CH; C2), 44.1 (CH; CHCH₃), 31.8 (CH₃; C6'), 28.8 (CH₂; CH₂CH₂CH₃), 23.6 (CH₃; C6), 20.7 (CH₂; CH₂CH₂CH₃), 14.2 (CH₃; CH₂CH₂CH₃), 14.0 (CH₃; CH*C*H₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{v} = 2069.0$ (10), 1969.4 (5), 1931.5 (100), 1927.1 (95), 1912.8 (40) [v(C=O)]; MS (70 eV): m/z for ¹⁸⁴W (%): 501 (24) [M]⁺, 473 (15) [M - CO]⁺, 445 (40) [M - 2CO]⁺; 415 (100).

4.7.2. Data for 13d

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.84 (s, br, 1H; 5-H), 3.07 and 2.94 (AB system, ${}^{2}J(H,H) = 22.7$ Hz, 2H; 3-H₂), 2.46 [s, 3H; 6-H₃], 1.97 [s, 3H; 6'-H₃], 2.05 (m, 2H; $CH_2CH_2CH_3$), 2.03 ("d", ³J(H,H) = 1.6 Hz, br, 3H; CHCH₃), 1.48 (m, 2H; CH₂CH₂CH₃), 0.95 (t, ${}^{3}J$ (H,H) = 7.3 Hz, 3H; CH₂CH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.9 and 198.5 [each C_q, 1:4, trans- and cis-CO; $W(CO)_5$], 183.3 (C_q; C=N), 151.6 (C_q; C1), 143.7 (C_q; CCH₃), 125.3 (CH; C5), 124.0 (C_q; C2), 44.0 (CH₂; C3), 31.7 (CH₃; C6'), 29.5 (CH₂; CH₂CH₂CH₂CH₃), 23.5 (CH₃; C6), 22.0 (CH₂; CH₂CH₂CH₃), 16.3 (CH₃; CHCH₃), 14.6 (CH₃; CH₂CH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{v} =$ 2068.5 (10), 1969.0 (3), 1931.1 (100), 1926.4 (95), 1911.1 (40) $[v(C \equiv O)]$; HRMS (ESI) calcd for C₁₇H₁₉NO5WNa $[M + Na]^+$: 524.0668; found: 524.0662; HRMS (ESI) calcd for $C_{15}H_{19}NO_3W$ [*M* - 2CO - H]⁻: 444.0793; found: 444.0803.

4.7.3. Data for 14d

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.94 (s, br, 1H; 3-H), 3.42 and 2.76 (AB system, ²*J*(H,H) = 22.7 Hz, 2H; 5-H₂), 2.46 [s, 3H; 6-H₃], 1.96 [s, 3H; 6'-H₃], for data of CC*H*₃ and *n*-propyl see **13d** due to strong overlap; ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.8 and 198.4 [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 184.2 (C_q; C=N), 150.3 (C_q; C1), 137.3 (C_q; CCH₃), 129.0 (C_q; C2), 127.3 (CH; C3), 44.7 (CH₂; C5), 32.0 (CH₃; C6'), 28.8 (CH₂; CH₂CH₂CH₃), 23.7 (CH₃; C6), 22.8 (CH₂; CH₂CH₂CH₃), 15.9 (CH₃; CH*C*H₃), 14.5 (CH₃; CH₂CH₂CH₃); for IR and MS data see **13d**.

4.8. Pentacarbonyl[cyclohexylidene-(3-methyl-5-propylcyclopenta-1,4-dienyl)- amine, N-W]tungsten(0) (12e), pentacarbonyl[cyclohexylidene-(4-methyl-2-propylcyclopenta-1,4-dienyl)- amine, N-W]tungsten(0) (13e), pentacarbonyl[cyclohexylidene-(4-methyl-2-propylcyclopenta-1,3-dienyl)- amine, N-W]tungsten(0) (14e)

(*E*)-*N*-Cyclohexyl-but-2-ene-1-carboximidoyl chloride (**5e**), generated from (*E*)-but-2-enonic acid cyclohexylamide (**4e**, 334 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) as described above, was added to pentacarbonyl[1-(ethylsulfanyl)but-1-ylidene]tungsten(0) (**1a**, 454 mg, 1.0 mmol) in dry dichloromethane (3 mL). Workup as described above afforded a bright yellow 10:1 mixture of compounds **12e** and **13e** (186 mg, 34%, $R_f = 0.7$ in 10:1 *n*-pentane/diethyl ether). Compound **12e** in CDCl₃ is slowly transformed into a 10:1 mixture of compounds **13e** and **14e**.



Data for 12e, 13e and 14e (NMR and MS data of a 3:10:1 mixture of compounds 12e, 13e and 14e).

4.8.1. Data for 12e

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.13 (m, 1H; 4-H), 5.61 (m, 1H; 2-H), 3.20 (m, 1H; CHCH₃), 2.90 (m, 2H; 6-H₂), 2.32 (m, 2H; 6'-H₂), 2.16 and 1.80 (each m, each 1H; CH₂CH₂CH₃), 1.61 (m, 2H; CH₂CH₂CH₃), 1.19 (d, ${}^{3}J(H,H) = 7.7$ Hz, 3H; CHCH₃), 0.99 (t, ${}^{3}J(H,H) = 7.4$ Hz, 3H; CH₂CH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.2 and 198.7 [each Cq, 1:4, trans- and cis-CO; W(CO)₅], 189.2 (C_q; C=N), 155.9 (C_q; C1), 141.2 (C_q; C5), 135.7 (CH; C4), 122.2 (CH; C2), 44.0 (CH; CHCH₃), 42.0 (CH₂; C6), 33.2 (CH₃; C6'), 28.8 (CH₂; CH₂CH₂CH₃), 27.7 and 27.5 (CH₂; C7 and C7'), 25.0 (CH₂; C8), 20.8 (CH₂; CH₂CH₂CH₃), 14.2 (CH₃; CH₂CH₂CH₃), 14.1 (CH₃; CH*C*H₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2068.5$ (10), 1968.7 (5), 1930.7 (100), 1926.6 (95), 1911.2 (40) [v(C=O)]; HRMS (ESI) calcd for $C_{20}H_{23}NO_5WNa$ $[M + Na]^+$: 564.0981; found: 564.0972; HRMS (ESI) calcd for C₁₈H₂₂- $NO_3W [M - 2CO - H]^-: 484.1160;$ found: 484.1161.

4.8.2. Data for 13e

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.84 (s, br, 1H; 5-H), 3.07 and 2.95 (AB system, ${}^{2}J(H,H) = 22.8$ Hz, 2H; 3-H₂), 2.83 [m, 2H; 6-H₂], 2.32 [s, 2H; 6'-H₂], 2.13 and 2.00 (m, 2H; $CH_2CH_2CH_3$), 2.03 ("d", ³J(H,H) = 1.5 Hz, br, 3H; CHCH₃), 1.89 and 1.66 (each m, 2:4 H; 7-, 7'- and 8-CH₂), 1.51 and 1.38 (each m, each 1H; CH₂CH₂CH₃), 0.94 $(t, {}^{3}J(H,H) = 7.3 \text{ Hz}, 3H; CH_2CH_2CH_3); {}^{13}C \text{ NMR}$ (100 MHz, CDCl₃, 300 K): δ 203.0 and 198.5 [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 188.9 (C_q; C=N), 150.7 (C_q; C1), 143.3 (C_q; CCH₃), 126.0 (CH; C5), 124.2 (C_q; C2), 44.1 (CH₂; C3), 42.0 (CH₂; C6), 32.8 (CH₂; C6'), 29.5 (CH₂; CH₂CH₂CH₃), 27.8 and 27.6 (CH₂; C7 and C7'), 22.2 (CH₂; CH₂CH₂CH₃), 16.3 (CH₃; CHCH₃), 14.6 (CH₃; CH₂CH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2067.7$ (10), 1968.2 (5), 1930.5 (100), 1925.8 (95), 1909.5 (40) $[v(C \equiv O)];$ HRMS (ESI) calcd for C₂₀H₂₃NO₅WNa $[M + Na]^+$: 564.0981; found: 564.0977; HRMS (ESI) calcd

for $C_{18}H_{22}NO_3W$ [*M* – 2CO – H]⁻: 484.1106; found: 484.1120; elemental analysis (%) calcd for $C_{20}H_{23}NO_5W$ (541.3): C, 44.38; H, 4.28; N, 2.59; found: C, 44.49; H, 4.18; N, 2.50%.

4.8.3. Data for 14e

¹H NMR (400 MHz, CDCl₃, 300 K): δ 5.94 (s, br, 1H; 3-H), 3.44 and 2.69 (AB system, ²*J*(H,H) = 22.6 Hz, 2H; 5-H₂), for data of CC*H*₃, *n*-propyl and cyclohexyl see **13e** due to strong overlap; ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 202.9 and 198.4 [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 189.9 (C_q; C=N), 149.4 (C_q; C1), 137.6 (C_q; CCH₃), 129.1 (C_q; C2), 127.4 (CH; C3), 45.3 (CH₂; C5), 42.3 (CH₃; C6'), 33.0 (CH₃; C6), 28.8 (CH₂; *C*H₂CH₂CH₃), 27.7 and 27.0 (each CH₂; C7 and C7'), 25.0 (CH₂; C8), 21.1 (CH₂; CH₂CH₂CH₃), 15.9 (CH₃; CH*C*H₃), 14.8 (CH₃; CH₂CH₂CH₃); for IR and MS data see **13e**.

4.9. (E)-N-Isopropyl-3-phenyl-acrylimidoyl chloride (5a), pentacarbonyl[(3-ethylsulfanyl-4-phenyl-cyclopent-2enylidene)-isopropyl- amine, N-W]tungsten(0) [(1 Z)-19a and (1E)-19a], pentacarbonyl[3-(isopropylamino)-5phenyl-penta-1,2,4-triene-1- ylidene]tungsten(0) (20a), (E)-N-isopropyl-3-phenyl-thioacrylimidic acid ethyl ester (21a)

(E)-N-Isopropyl-3-phenyl-acrylimidoyl chloride (5a), generated as described above from (2E)-N-isopropyl-3-phenyl acrylamide (4a, 378 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol), was reacted with pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (1b, 412 mg, 1.0 mmol). Work-up gave a colorless compound 21a (23 mg, 10%, $R_f = 0.5$ in 10:1 *n*-pentane/diethyl ether) and a thermolabile red compound ($R_{\rm f} = 0.4$ in 10:1 *n*-pentane/ diethyl ether) which was transformed into compound (Z)-**19a** (310 mg, 53%, $R_f = 0.7$ in 10:1 *n*-pentane/diethyl ether, yellow oil) within 30 min. Equilibration of (Z)-19a and (E)-**19a** ($R_{\rm f} = 0.5$ in 10:1 *n*-pentane/diethyl ether) is accompanied by strong decomposition. ¹H NMR data of (E)-19a were collected from a sample freshly chromatographed and very dilute. A more polar dark red to violet fraction contained compound **20a** (64 mg, 12%, $R_{\rm f} = 0.5$ in 1:1:1 *n*pentane/diethyl ether/dichloromethane, violet oil).



4.9.1. Data for **5a** (collected from a 1:1 mixture of amid and phosphorous oxychloride in CDCl₃ after completion of the reaction at 20 °C, 6 h)

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.99 (d, ³*J*(H,H) = 15.2 Hz, 1H; PhC*H*), 7.77 (d, ³*J*(H,H) = 15.2 Hz, 1H; PhCH=C*H*), 7.77 (m, 2H; *o*-CH Ph), 7.54 (m, 1H; *p*-CH Ph), 7.46 (m, 2H; *m*-CH Ph), 4.54 (sept., ³*J*(H,H) = 6.6 Hz;

2H; NCH), 1.55 [d, ${}^{3}J(H,H) = 6.6$ Hz, 6H; NCH(CH₃)₂]; (1³C NMR (100 MHz, CDCl₃, 300 K): δ 165.7 (C_q; C=N), (154.4 (CH; PhCH), 133.2 (CH; *p*-C Ph), 132.2 (C_q; *i*-C Ph), 129.9 (CH; *o*-C Ph), 129.1 (CH; *m*-C Ph), 117.3 (CH; O

PhCH=*C*H), 53.5 (CH; NCH), 20.6 [NCH(*C*H₃)₂].



4.9.2. Data for (Z)-19a{(E)-19a} (obtained from a 2:3 Z/E mixture)

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.36 {7.36} (m, 2H; m-CH Ph), 7.30 {7.30} (m, 1 H; p-CH Ph), 7.13 $\{7.13\}$ (m, 2H; o-CH Ph), 6.89 $\{6.32\}$ (d, ${}^{4}J(H,H) = 1.4$ Hz {1.3 Hz}, 1H; 2-H), 4.08 {3.98} (m, 1H; 4-H), 3.82 {4.36} (sept. {br}, ${}^{3}J(H,H) = 6.6$ Hz, 1H; NCH), 3.35 {3.45} (dd. ${}^{2}J(H,H) = 17.7 \text{ Hz} \{17.8 \text{ Hz}\}, {}^{3}J(H,H) = 7.6 \text{ Hz}$ $\{7.5 \text{ Hz}\}, 1\text{H}; cis-5-\text{H}_2\}, 2.96 \{2.86\} (q, {}^{3}J(\text{H},\text{H}) = 7.5 \text{ Hz}\}$ {7.5 Hz}, 2H; $SCH_2CH_3),$ 2.77 {2.99} (dd, $^{2}J(H,H) = 17.7 \text{ Hz} \{17.8 \text{ Hz}\}, \,^{3}J(H,H) = 2.6 \text{ Hz} \{2.4 \text{ Hz}\},$ 1H; trans-5-H₂), 1.38 {1.32} (t, ${}^{3}J(H,H) = 7.5$ Hz, 3H; SCH_2CH_3), 1.29 and 1.27 {1.41 and 1.40} [each d, ${}^{3}J(H,H) = 6.6 \text{ Hz} \{6.6 \text{ Hz}\}, 3H; \text{ NCH}(CH_{3})_{2}]; {}^{13}C \text{ NMR}$ (100 MHz, CDCl₃, 300 K): δ 201.7 and 199.4 [C_q, 1:4, trans- and cis-CO; W(CO)₅], 182.5 (C_a; C=N), 172.5 (C_a; C3), 141.0 (Cq; i-C Ph), 129.1 (CH; m-C Ph), 128.2 (CH; C2), 127.8 (CH; p-C Ph), 127.1 (CH; o-C Ph), 60.4 (CH; NCH), 52.1 (CH; C4), 41.9 (CH₂; C5), 27.1 (SCH₂CH₃), 23.8 and 23.7 [NCH(CH₃)₂], 13.3 (SCH₂CH₃); IR (cyclohexane) $[\text{cm}^{-1} (\%)]: \tilde{v} = 2065.7 (30), 1962.4 (5), 1927.1$ (100), 1918.3 (95), 1908.0 (95) [v(C=O)], 1569.7 (10), 1545.4 (10) [v(C=C) and v(C=N)]; IR (diffuse reflexion) $[\text{cm}^{-1} (\%)]: \tilde{v} = 2064.4 (10), 1965.8 (4), 1875.5 (100),$ 1568.3 (7), 1544.0 (15), 1494.5 (1), 1454.0 (3), 1304.5 (1), 1172.8 (1); HRMS (ESI) calcd for C₂₁H₂₁NSO₅WNa $[M + Na]^+$: 606.0543; found: 606.0513.



4.9.3. Data for 20a

¹H NMR (400 MHz, CDCl₃, 300 K): δ 9.52 (s, br, 1H; NH), 7.98 (d, ³*J*(H,H) = 15.3 Hz, 1H; PhC*H*), 7.53 (m, 2H; *o*-CH Ph), 7.34 (m, 1H; *p*-CH Ph), 7.31 (m, 2H; *m*-CH Ph), 6.95 (m, ³*J*(H,H) = 15.3 Hz, 1H; PhCH=C*H*), 4.62 (m, 1H; NCH), 1.40 (d, ³*J*(H,H) = 6.6 Hz, 6H; NCH(CH₃)₂); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.3 and 197.4 [C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅],

(C_q; W=C),[25] 151.6 (C_q; C4), 150.1 (CH; C6), 134.3 (C_q; *i*-C Ph), 131.2 (CH; *p*-CH Ph), 129.1 (CH; *m*-CH Ph), 128.7 (CH; *o*-CH Ph), 121.8 (CH; C5), 112.2 (C_q; C3), 50.3 (NCH), 21.7 (NCH(CH₃)₂); IR (dichloromethane) [cm⁻¹ (%)]: $\tilde{v} = 3365.9$ (5) [v(N–H)], 2082.8 (1), 1993.7 (25), 1929.4 (100) [v(C=O)], 1625.9 (10), 1540.1 (10), 1467.1 (10), 1452.5 (10); HRMS (ESI) calcd for C₁₉H₁₅NO₅WNa [M + Na]⁺: 544.0355; found: 544.0340.



4.9.4. Data for anti-21a{syn-21a} (obtained from a 1:1 mixture)

¹H NMR (400 MHz, CDCl₃, 300 K) δ 7.47 {7.47} (each m, 1H; p-CH Ph), 7.34 {7.34} (each m, 2H; m-CH Ph), 7.34 {7.34} (each m, 2H; o-CH Ph), 7.22 {7.27} (each d, ${}^{3}J(H,H) = 16.2 \text{ Hz} \{15.8 \text{ Hz}\}, 1H; PhC-H), 6.99 \{6.81\}$ [each d, ${}^{3}J(H,H) = 16.2 \text{ Hz} \{15.8 \text{ Hz}\}, 1H; N=CC-H,$ NOE (+) with NCH(CH₃)₂ {no NOE with NCH(CH₃)₂}], 4.11 {4.03} [each sept., ${}^{3}J(H,H) = 6.2$ Hz, 1H; NCH(CH₃)₂, NOE (+) with N=CC-H {no NOE with N=CC-H}], 2.93 $\{2.96\}$ (each q, ${}^{3}J(H,H) = 7.4$ Hz, 2H; SCH₂), 1.29 $\{1.28\}$ $(t, {}^{3}J(H,H) = 7.4 \text{ Hz}, \text{ each } 3H; \text{ SCH}_{2}CH_{3}), 1.22 \{1.17\}$ [each d, ${}^{3}J(H,H) = 6.2$ Hz, 6 H; NCH(CH₃)₂]; ${}^{13}C$ NMR (100 MHz, CDCl₃, 300 K): δ 158.0 {157.7} (C_a; C=N), 136.8 {136.7} (CH; Ph-CH), 136.7 {136.0} (C_q; *i*-C Ph), 129.0 {128.9} (CH; p-CH Ph), 128.6 {128.7} (CH; m-CH Ph), 127.3 {127.3} (CH; o-CH Ph), 120.0 {126.5} (CH; Ph-CH=CH), 53.8 {51.6} [NCH(CH₃)₂], 23.7 {27.0} (SCH₂), 24.1 {23.2} [NCH(CH₃)₂], 14.3 {15.7} (SCH₂CH₃). IR (cyclohexane) $[\text{cm}^{-1} (\%)]: \tilde{\nu} = 1630.0 (50), 1587 (100)$ [v(C=O)]; MS (70 eV): m/z for ¹⁸⁴W (%): 233.1 (5) $[M]^+$, 172.1 (35) $[M - SEt]^+$, 130.0 (100) $[M - SEt - C_3H_6]^+$; HRMS (ESI) calcd for $C_{14}H_{19}NS [M + H]^+$: 234.1313; found: 234.1311.

4.10. (2s-cis, 3Z, 4s-cis,5E)-1,1,1,1,1-Pentacarbonyl-2ethylsulfanyl-6-phenyl-4-N-cyclohexylamino-1-tungsten-1,3,5-hexatriene (18b), pentacarbonyl[(3-ethylsulfanyl-4phenyl-cyclopent-2-enylidene)- cyclohexyl-amine, N-W]tungsten(0) [(1Z)-19b and (1E)-19b], pentacarbonyl[3-(cyclohexylamino)-5-phenyl-penta-1,2,4triene-1-ylidene]tungsten(0) (20b)

(E)-N-Cyclohexyl-3-phenyl-acrylimidoyl chloride (**5b**), generated from (2E)-N-cyclohexyl-3-phenyl acrylamide (**4b**, 378 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol), was reacted with pentacarbonyl-

[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 412 mg, 1.0 mmol). Work-up afforded red crystals of compound **18b** (225 mg, 36%, $R_{\rm f} = 0.3$ in 10:1 *n*-pentane/diethyl ether). Compound **18b** was transformed into (Z)-**19b** ($R_f = 0.7$ in 10:1 *n*-pentane/diethyl ether, yellow oil) in CDCl₃, 20 °C, within 3 h. Equilibration of (Z)-**19b** and (E)-**19b** in solution was accompanied by strong decomposition. A more polar dark red to violet fraction yielded compound **20b** (95 mg, 17%, $R_{\rm f} = 0.3$ in *n*-pentane/dichloromethane 1:1, violet oil).



4.10.1. Spectroscopic data of 18b

¹H NMR (600 MHz, CDCl₃, 233 K): δ 10.06 (d, br, ${}^{3}J(H,H) = 8.1 \text{ Hz } 1\text{H}; \text{ NH}), 7.63 \text{ (m, 2H; } o\text{-CH Ph}), 7.57$ $(dd, {}^{3}J(H,H) = 15.9 \text{ Hz}, {}^{3}J(H,H) = 3.1 \text{ Hz}, 1H; CH-Ph),$ 7.51 (m, 3H; m/p-CH Ph), 7.00 (d, ${}^{3}J(H,H) = 15.9$ Hz, 1H; CHCHPh), 6.86 (s, 1H; 3-H), 3.75 (m, 1H; NCH), 2.94 (q, 2H; SCH₂CH₃), 2.15, 1.90, 1.73, 1.65, 1.37 (each m, 2:2:1:3:2H; CH₂-cyclohexyl), 1.33 (t, 3H; SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃, 233 K): δ 225.4 (C_a; W=C), 202.9 and 198.7 [C_q, 1:4, trans- and cis-CO; W(CO)₅], 160.4 (C_q; C4), 143.0 (CH; C6), 134.0 (C_q; *i*-C Ph), 130.8 (CH; p-C Ph), 129.1 (CH; m-C Ph), 127.9 (CH; o-C Ph), 119.9 (CH; C5), 115.1 (CH; C3), 54.9 (CH; NCH), 32.5 (SCH₂CH₃), 32.1 (CH₂; NCHCH₂), 24.7 (NCHCH₂CH₂), 24.5 (NCHCH₂CH₂CH₂), 11.4 (SCH₂*C*H₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2060.6$ (40), 1934.9 (100), 1918.4 (70), 1907.9 (40) [v(C=O)]; IR (diffuse reflexion) $[\text{cm}^{-1} (\%)]$: $\tilde{v} = 2056.6$ (20), 1882.2 (100), 1627.7 (5), 1578.7 (5), 1505.5 (20), 1452.5 (3), 1387.9 (3), 1363.6 (4), 1325.6 (4), 1310.8 (5), 1288.1 (7), 1257.3 (4), 1233.6 (4); HRMS (ESI) calcd for $C_{24}H_{25}NSO_5WNa [M + Na]^+: 646.0857;$ found: 646.0842; elemental analysis (%) calcd for C₂₄H₂₅NSO₅W (623.4): C, 46.24; H, 4.04; N, 2.25; found C, 46.15; H, 3.82; N, 2.18%.

4.10.2. Molecular structure analysis of 18b (code 3657.AUM)

Formula $C_{24}H_{25}NO_5SW$, $M_r = 623.36 \text{ g mol}^{-1}$, red crystal, $0.45 \times 0.25 \times 0.20 \text{ mm}$, a = 15.684(1), b = 8.513(1), c = 18.472(1) Å, $\beta = 97.89(1)^\circ$, V = 2443.0(4) Å³, $\rho_{calcd} = 1.695 \text{ g cm}^{-3}$, $\mu = 48.47 \text{ cm}^{-1}$, empirical absorption correction ($0.219 \le T \le 0.444$), Z = 4, monoclinic, space group P_{21}/n (no. 14), $\lambda = 0.71073$ Å, T = 198 K, ω and φ scans, 15269 reflections collected ($\pm h$, $\pm k$, $\pm l$), [($\sin\theta$)/ λ]_{max} = 0.67 Å⁻¹, 5861 independent ($R_{int} = 0.029$) and 5106 observed reflections [$I \ge 2\sigma(I)$], 295 refined parameters, R = 0.023, $wR_2 = 0.051$, max./min. residual electron

density 1.22/-0.93 e Å⁻³, hydrogen atom at N1 from difference fourier map, others calculated and refined as riding atoms [24].



4.10.3. Data for (Z)-19b

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.36 (m, 2H; m-CH Ph), 7.30 (m, 1H; p-CH Ph), 7.13 (m, 2H; o-CH Ph), 6.91 (s, 1H; 2-H), 4.07 (m, 1H; Ph-CH), 3.34 (dd, ${}^{2}J(\text{H}_{2}\text{H}) = 17.7 \text{ Hz}, {}^{3}J(\text{H},\text{H}) = 7.5 \text{ Hz}, 1\text{H}; cis-5-\text{H}_{2}), 2.95$ $(q, {}^{3}J(H,H) = 7.5 \text{ Hz}, 2H; \text{ SC}H_2CH_3), 3.30 (m, 1H;$ NCH), 2.75 (dd, ${}^{2}J(H,H) = 17.7$ Hz, ${}^{3}J(H,H) = 2.6$ Hz, 1H; *trans*-5-H₂), 1.87,1.66 and 1.29 (each m, 4:4:2H; CH₂-cyclohexyl), 1.37 (t, ${}^{3}J$ (H,H) = 7.5 Hz, 3H; SCH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 201.8 and 199.5 [each Cq, 1:4, trans- and cis-CO; W(CO)₅], 182.6 (C_q; C=N), 172.4 (C_q; C3), 140.9 (C_q; *i*-C Ph), 129.1 (CH; m-C Ph), 128.2 (CH; C2), 127.7 (CH; p-C Ph), 127.1 (CH; o-C Ph), 69.1 (CH; NCH), 52.0 (CH; C4), 41.9 (CH₂; C5), 34.1 (br), 25.8, 25.2, 24.9 (each m, 2:2:1; CH₂-cyclohexyl), 27.0 (SCH₂CH₃), 13.2 (SCH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]: \tilde{v} = 2065.2 (30), 1961.5 (5),$ 1927.9 (100), 1918.4 (95), 1915.3 (95), 1907.3 (100) $[v(C \equiv O)]$, 1567.6 (10), 1545.7 (15) [v(C = C)] and v(C = N); IR (diffuse reflexion) $[cm^{-1} (\%)]$: $\tilde{v} = 2928.0$ (5), 2857.0 (2), 2064.0 (10), 1965.0 (4), 1889.9 (100), 1567.5 (10), 1541.0 (20), 1453.2 (4); HRMS (ESI) calcd for $C_{24}H_{25}NSO_5WNa [M + Na]^+$: 646.0856; found: 646.0859.



4.10.4. Data for 20b

¹H NMR (400 MHz, CDCl₃, 300 K): δ 9.24 (br, 1H; NH), 7.99 (d, ³*J*(H,H) = 15.4 Hz, 1H; CH=C*H*Ph), 7.45 (m, 2H; *o*-CH Ph), 7.31 (m, 3H; *m*-and *p*-CH Ph), 6.88 (d, ³*J*(H,H) = 15.4 Hz, 1H; C*H*=CHPh), 4.32 (m, 1H; NCH), 2.18 and 1.41 (each m, 2H; NCHC*H*₂), 1.82 and 1.42 (each m, 2H; NCHCH₂C*H*₂), 1.63 (m, 2H; NCHCH₂CH₂CH₂C*H*₂); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.4 and 197.4 [C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 193.4 (C_q, small/broad; W=C), 151.2 (C_q; C4), 150.1 (CH; CH=CHPh), 134.2 (C_q; *i*-C Ph), 131.3 (CH; *p*-CH Ph), 129.1 (CH; *m*-CH Ph), 128.6 (CH; *o*-CH Ph), 121.7 (CH; CH=CHPh), 112.3 (C_q, small/broad; C3), 57.3 (CH; NCH), 32.0 (CH₂; NCH*C*H₂), 25.1 (CH₂; NCHCH₂CH₂CH₂), 24.5 (CH₂; NCHCH₂CH₂); IR (dichloromethane) [cm⁻¹ (%)]: $\tilde{v} = 3363.9$ (5) [v(N–H)], 2082.8 (2), 1994.2 (30), 1929.1 (100) [v(C \equiv O)], 1625.7 (10), 1539.1 (10); IR (diffuse reflexion) [cm⁻¹ (%)]: $\tilde{v} = 2082.6$ (0.5), 1995.9 (10), 1898.1 (100) [v(C \equiv O)]; HRMS (ESI) calcd for C₂₂H₁₉NO₅WNa [M + Na]⁺: 584.0668; found: 584.0683.

4.11. (2s-cis,3Z,4s-cis,5E)-1,1,1,1,1-Pentacarbonyl-2ethylsulfanyl-6-methyl-4-N-isopropylamino-1-tungsten-1,3,5-hexatriene (**18d**), pentacarbonyl[(3-ethylsulfanyl-4methyl-cyclopent-2-enylidene)-isopropyl-amine, N-W]tungsten(0) [(1Z)-**19d**]

(E)-N-Isopropyl-but-2-ene-1-carboximidoyl chloride (5d), generated as described above from (E)-but-2-enonic acid isopropylamide (4d, 254 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) was reacted with pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (1b, 412 mg, 1.0 mmol). Addition of triethylamine (808 mg, 8.0 mmol) at -78 °C and direct transfer of the mixture to a precooled column at -78 °C on silica gel (15×2 cm, 1:1 *n*-pentane/diethyl ether) afforded a red fraction, from which red crystals of compound 18d were obtained at -20 °C from *n*-pentane (110 mg, 21%, $R_{\rm f} = 0.5$ in 10:1 *n*pentane/diethyl ether). In solution (CDCl₃, 20 °C) 18d was transformed into (Z)-19d ($R_{\rm f} = 0.6$ in 10:1 *n*-pentane/ diethyl ether, yellow oil) within 3 h. Equilibration of (Z)-**19d** and (*E*)-**19d** ($R_f = 0.5$ in 10:1 *n*-pentane/diethyl ether) is accompanied by strong decomposition.



4.11.1. Data for 18d

¹H NMR (600 MHz, CDCl₃, 233 K): δ 10.00 (s, br, ${}^{3}J(H,H) = 8.3$ Hz, 1H; NH), 6.89 (dq, ${}^{3}J(H,H) = 15.3$ Hz, ${}^{3}J(H,H) = 6.9$ Hz, 1H; 6-H), 6.72 (s, 1H; 3-H), 6.40 (dq, ${}^{3}J(H,H) = 15.3 \text{ Hz}, {}^{4}J(H,H) = 1.2 \text{ Hz}, 1H; 5-H), 4.10$ ${}^{3}J(H,H) = 8.3 \text{ Hz}, \quad {}^{3}J(H,H) = 6.9 \text{ Hz},$ [d{sept}, 1H: $CH(CH_3)_2$], 2.90 (q, ${}^{3}J(H,H) = 7.5$ Hz, 2H; SCH_2CH_3), 2.07 (dd, ${}^{3}J(H,H) = 6.9$ Hz, ${}^{4}J(H,H) = 1.2$ Hz, 3H; 7-CH₃), 1.47 [d, ${}^{3}J(H,H) = 6.6$ Hz, 6 H; CH(CH₃)₂], 1.32 (t, 3H; SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃, 233 K): δ 224.2 (C_q; W=C), 202.9 and 198.7 [C_q, 1:4, trans- and cis-CO; W(CO)₅], 160.8 (C_q; C4), 143.7 (CH; C6), 124.0 (CH; C5), 115.1 (CH; C3), 47.3 (CH; NCH), 32.3 (SCH₂CH₃), 22.0 (CH₃; NCH(CH₃)₂]), 19.9 (CH₃; C7), 11.4 (SCH₂CH₃); IR (cyclohexane) $[\text{cm}^{-1} (\%)]$: $\tilde{v} = 3201.1$ (1) [v(N-H)], 2061.0 (30), 1935.1 (100), 1918.9 (70), 1906.8 (50) [v(C=O)], 1556.0 (5), 1508.4 (5); HRMS (ESI) calcd for $C_{16}H_{19}NSO_5W[M-H]^-$: 520.0410; found: 520.0415; elemental analysis (%) calcd for $C_{16}H_{19}NSO_5W$ (521.2): C, 36.87; H, 3.67; N, 2.69; found C, 36.79; H, 3.69; N, 2.61%.



4.11.2. Data for (Z)-19d

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.66 (d, ${}^{3}J(H,H) = 1.2$ Hz, 1 H; 2-H), 3.83 (sept., ${}^{3}J(H,H) = 6.6$ Hz, 1H; NCH), 3.08 (dd, ${}^{2}J(H,H) = 17.1 \text{ Hz}, {}^{3}J(H,H) = 7.0 \text{ Hz},$ 1H; cis-5-H₂), 3.02 (m, 1H; CHCH₃), 2.98 (q, ${}^{3}J(H,H) =$ 7.4 Hz, 2H; SCH₂CH₃), 2.38 (dd, ${}^{2}J(H,H) = 17.1$ Hz, ${}^{3}J(H,H) = 2.0$ Hz, 1H; trans-5-H₂), 1.43 (t, ${}^{3}J(H,H) =$ 7.4 Hz, 3H; SCH₂CH₃), 1.28 and 1.27 [each d, ${}^{3}J(H,H) =$ 6.6 Hz, 3H; NCH(CH₃)₂], 1.24 (d, ${}^{3}J(H,H) = 7.0$ Hz, 3H; CHCH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 201.7 and 199.4 [each Cq, 1:4, trans- and cis-CO; W(CO)5], 182.5 (C_q; C=N), 174.6 (C_q; C3), 126.7 (CH; C2), 60.1 (CH; NCH), 40.9 (CH; C4), 40.2 (CH₂; C5), 26.9 (SCH₂CH₃), 23.8 and 23.7 (each CH₃; NCH(CH₃)₂]), 20.6 (CH₃; CHCH₃), 13.4 (SCH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2065.2$ (25), 1928.9 (70), 1922.5 (60), 1915.4 (55), 1908.3 (100) [v(C=O)], 1569.7 (5), 1545.5 (5) [v(C=N) andv(C=C); HRMS (ESI) calcd for C₁₆H₁₈NSO₅W [M - H]⁻: 520.0410; found: 520.0429.

4.12. (2s-cis,3Z,4s-cis,5E)-1,1,1,1,1-Pentacarbonyl-2ethylsulfanyl-6-phenyl-4-N-methylamino-1-tungsten-1,3,5hexatriene (**18f**), pentacarbonyl[(3-ethylsulfanyl-4-phenylcyclopent-2-enylidene)-methyl- amine, N-W]tungsten(0) [(1Z)-**19f** and (1E)-**19f**], pentacarbonyl[3-(methylamino)-5-phenyl-penta-1,2,4-triene-1-ylidene]tungsten(0) (**20f**)

(*E*)-*N*-Methyl-3-phenyl-acrylimidoyl chloride (**5f**), generated *in situ* from (2*E*)-*N*-methyl-3-phenyl acrylamide (**4f**, 332 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol), was reacted with pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 412 mg, 1.0 mmol) as described above at -78 °C to give compound **18f** (250 mg, 49%, $R_f = 0.6$ in 1:2 *n*-pentane/diethyl ether, red oil). Compound **18f** in CDCl₃, 20 °C was transformed into the yellow compound (*Z*)-**19f** (195 mg, 35%, $R_f = 0.6$ in 10:1 *n*-pentane/diethyl ether, yellow oil) after short time. An equilibrium between (*Z*)-**19f** and (*E*)-**19f** ($R_f = 0.4$ in 10:1 *n*-pentane/diethyl ether) with a *E*/*Z*-ratio of 2:3 was established after 14 d, 20 °C. A more polar dark red to violet fraction yielded compound **20f** (109 mg, 22%, $R_f = 0.6$ in 1:1 dichloromethane/diethyl ether, violet oil).



4.12.1. Data for 18f

¹H NMR (600 MHz, CDCl₃, 233 K): δ 0.02 (q, br, ³*J*(H,H) = 5.6 Hz, 1H; NH), 7.65 (m, 2 H; *o*-CH Ph), 7.65 (d, ³*J*(H,H) = 16.1 Hz, 1H; C*H*-Ph), 7.51 (m, 3H; *m*/ *p*-CH Ph), 6.96 (d, ³*J*(H,H) = 16.1 Hz, 1H; C*H*CHPh), 6.86 (s, 1H; 3-H), 3.32 (d, ³*J*(H,H) = 5.6 Hz, 3 H; NCH₃), 2.95 (q, br, 2H; SCH₂CH₃), 1.34 (t, 3H; SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃, 233 K): δ 225.4 (C_q; W=C), 202.9 and 198.5 [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 163.3 (C_q; C4), 144.6 (CH; C6), 133.8 (C_q; *i*-C Ph), 131.2 (CH; *p*-C Ph), 129.1 and 128.2 (CH; *o*and *m*-C Ph), 118.7 (CH; C5), 114.9 (CH; C3), 32.7 (SCH₂CH₃), 31.1 (CH₂; NCH₃), 11.4 (SCH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{v} = 2060.9$ (40), 1936.3 (100), 1911.8 (60) [v(C=O)]; HRMS (ESI) calcd for C₁₉H₁₇NSO₅WNa [*M* + Na]⁺: 578.0229; found: 578.0231.



4.12.2. Data for (Z)-19f {(E)-19f} (obtained from a 10:6.5 Z/E mixture)

¹H NMR (600 MHz, CDCl₃, 298 K): δ 7.36 {7.36} (m, 2H; m-CH Ph), 7.31 {7.31} (m, 1H; p-CH Ph), 7.15 $\{7.15\}$ (m, 2H; o-CH Ph), 6.58 $\{6.26\}$ (d, ${}^{4}J$ (H,H) = 1.4 Hz $\{1.4 \text{ Hz}\}, 1\text{H}; 2\text{-H}, \text{ NOE } (+) \text{ with } \text{SC}H_2 \{\text{NOE } (+) \text{ with } \}$ NCH₃ and SCH₂}), 4.12 {4.05} (m, 1H; 4-H), 3.54 {3.68} (s, 3H; NCH₃, NOE (+) with 5-H₂ {NOE (+) with 2-H}), 3.24 {3.40} $(ddd,^2 J(H,H) = 18.0 \text{ Hz} \{18.0 \text{ Hz}\},$ ${}^{3}J(H,H) = 7.6 \text{ Hz} \{7.6 \text{ Hz}\}, {}^{5}J(H,H) = 0.7 \text{ Hz} \{1.4 \text{ Hz}\},$ 1H; cis-5-H₂, NOE (+) with 4-H, NCH₃ and trans-5-H₂ {NOE (+) with 4-H and *trans*-5-H₂}), 2.95 {2.87} (q, 2H; $^{2}J(H,H) = 18.0 \text{ Hz}$ SCH_2CH_3 , 2.64 {2.90} (ddd, {18.0 Hz}, ${}^{3}J(H,H) = 2.6$ Hz {2.6 Hz}, ${}^{5}J(H,H) = 0.7$ Hz {1.4 Hz}, 1H; trans-5-H₂), 1.38 {1.32} (t, 3H; SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃, 298 K): δ 202.5 and 199.0 {202.4 and 198.7} [each C_q, 1:4, trans- and cis-CO; $W(CO)_5$], 184.4 {184.0} (C_q; C=N), 173.1 {175.4} (C_q; C=N) C3), 140.7 {140.5} (C_a; *i*-C Ph), 129.1 {129.0} (CH; *m*-C Ph), 127.9 {127.8} (CH; p-C Ph), 127.2 {127.2} (CH; o-C Ph), 126.2 {115.1} (CH; C2), 53.4 {53.9} (NCH₃), 52.5 {51.3} (CH; C4), 40.8 {49.0} (CH₂; C5), 27.2 {26.9} (SCH₂CH₃), 13.2 {13.0} (SCH₂CH₃); IR (cyclohexane)

[cm⁻¹ (%)]: $\tilde{v} = 2066.4$ (20), 1965.0 (5), 1925.3 (100), 1910.9 (50) [v(C = O)], 1606.1 (5), 1544.1 (5) [v(C = C) and v(C = N)]; IR (diffuse reflexion) [cm⁻¹ (%)]: $\tilde{v} = 2065.6$ (5), 1971.0 (3), 1897.1 (100), 1602.4 (4), 1541.7 (10); MS (70 eV): m/z for ¹⁸⁴W (%): 555 (30) [M]⁺, 471 (75) [M – 3CO]⁺, 442 (50) [M – 3CO – Et]⁺, 415 (40) [M – 5CO]⁺, 231 (100) [M – W(CO)₅]⁺, 170 (70) [M – W(CO)₅ – SEt]⁺; HRMS (ESI) calcd for C₁₉H₁₇NSO₅WNa [M + Na]⁺: 578.0253; found: 578.0237.



4.12.3. Data for 20f

¹H NMR (400 MHz, CDCl₃, 300 K): δ 10.36 (s, br, 1H; NH), 7.98 (d, ³*J*(H,H) = 15.4 Hz, 1H; PhC*H*), 7.51 (m, 2H; *o*-CH Ph), 7.28 (m, 3H; *m*- and *p*-CH Ph), 6.98 (d, ³*J*(H,H) = 15.4 Hz, 1H; PhCH=C*H*), 3.38 (s, 3H; NCH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.3 and 197.4 [each C_q, 1:4, *trans*- and *cis*-CO; W(CO)₅], 193.9 (C_q, dynamically broadened; W=C), 154.1 (C_q; C4), 149.8 (CH; C6), 134.2 (C_q; *i*-C Ph), 131.2 (CH; *p*-CH Ph), 129.0 and 128.6 (each CH; *o*- and *m*-CH Ph), 121.9 (CH; C5), 112.3 (C_q; C3), 34.1 (NCH₃); IR (dichloromethane) [cm⁻¹ (%)]: $\tilde{v} = 3386.7$ (4) [v(N–H)], 2083.0 (1), 1992.9 (35), 1929.8 (100) [v(C=O)], 1625.5 (7), 1565.3 (6), 1473.1 (10), 1451.4 (3); MS-ES (ES⁻/NaBF₄-addition): *m/z* (%) = 580 (100) [*M* + BF₄]⁻, 1073 (10) [2*M* + BF₄]⁻.

4.13. (E)-N-Propyl-3-phenyl-acrylimidoyl chloride (**5**g), (2s-cis,3Z,4s-cis,5E)-1,1,1,1,1-Pentacarbonyl-2ethylsulfanyl-6-phenyl-4-N-propylamino-1-tungsten-1,3,5hexatriene (**18**g), pentacarbonyl[(3-ethylsulfanyl-4-phenylcyclopent-2-enylidene)-propyl-amine, N-W]tungsten(0) [(1Z)-**19**g and (1E)-**19**g], pentacarbonyl[3-(propylamino)-5-phenyl-penta-1,2,4-triene-1ylidene]tungsten(0) (**20**g)

(*E*)-*N*-Propyl-3-phenyl-acrylimidoyl chloride (**5g**), generated *in situ* as described above from (2*E*)-*N*-propyl-3-phenyl acrylamide (**4g**, 378 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) and pentacarbonyl-[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 412 mg, 1.0 mmol). Work-up gave red crystals of compound **18g** (328 mg, 56%, $R_f = 0.3$ in 7:3 *n*-pentane/diethyl ether). In solution (CDCl₃, 20 °C) **18g** was transformed into (*Z*)-**19g** ($R_f = 0.7$ in 10:1 *n*-pentane/diethyl ether, yellow oil) within 4 h. In CDCl₃ (*Z*)-**19g** and (*E*)-**19g** ($R_f = 0.5$ in 10:1 *n*-pentane/diethyl ether, yellow oil) had an equilibrium with a *E*/*Z*-ratio of 3:4 after 11 d at 20 °C. A more polar dark red to violet fraction yielded **20g** (160 mg, 31%, $R_f = 0.7$ in dichloromethane, violet oil).

4.13.1. Data for 5g (obtained from a 1:1 mixture of the amide and phosphorous oxychloride in CDCl₃ after 48 h, 20 °C; yield estimated according to NMR ca. 85%)

¹H NMR (400 MHz, CDCl₃, 300 K): δ 8.02 (d, ³*J*(H,H) = 15.3 Hz, 1 H; PhC*H*), 7.82 (d, ³*J*(H,H) = 15.3 Hz, 1H; PhCH=C*H*), 7.76 (m, 2H; *o*-CH Ph), 7.55 (m, 1H; *p*-CH Ph), 7.46 (m, 2H; *m*-CH Ph), 3.87 (t, ³*J*(H,H) = 7.3 Hz; 2 H; NCH₂), 1.95 (m, 2H; NCH₂CH₂), 1.08 (t, ³*J*(H,H) = 7.4 Hz, 3H; NCH₂CH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 167.8 (C_q; C=N), 154.5 (CH; Ph*C*H), 133.3 (CH; *p*-C Ph), 132.2 (C_q; *i*-C Ph), 130.0 (CH; *o*-C Ph), 129.1 (CH; *m*-C Ph), 117.1 (CH; *C*H=CH-Ph), 51.1 (CH₂; NCH₂), 20.9 (NCH₂*C*H₂CH₃), 10.9 (NCH₂CH₂CH₃).



4.13.2. Data for 18g

¹H NMR (400 MHz, CDCl₃, 300 K): δ 9.92 (s, br, 1H; NH), 7.56 (m, 2H; o-CH Ph), 7.50 (d, br, 1H; CH-Ph), 7.45 (m, 3H; *m/p*-CH Ph), 6.94 (d, br, 1H; CHCHPh), 6.85 (s, 1H; 3-H), 3.52 (m, 2H; NCH₂), 2.94 (q, 2 H; SCH₂CH₃), 1.91 (m, 2H; NCH₂CH₂), 1.34 (t, 3H; SCH₂CH₃), 1.08 (t, 3H; NCH₂CH₂CH₃); 13 C NMR (100 MHz, CDCl₃, 300 K): δ 229.6 (C_q; W=C), 202.8 and 199.0 [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 162.5 (C_q; C4), 143.4 (CH; C6), 134.5 (C_q; *i*-C Ph), 130.9 (CH; p-C Ph), 129.2 and 128.0 (CH; o- and m-C Ph), 120.2 (CH, br; C5), 115.6 (CH, br; C3), 47.0 (CH₂; NCH₂), 33.0 (SCH₂CH₃), 22.5 (NCH₂CH₂CH₃), 11.8 (SCH₂CH₃), 11.4 (NCH₂CH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2060.7$ (40), 1935.1 (100), 1917.7 (60) [$v(C \equiv O)$]; HRMS (ESI) calcd for $C_{21}H_{20}NSO_5W$ [M – H]⁻: 582.0567; found: 582.0593.



4.13.3. Data for (Z)-19g {(E)-19g} (obtained from a 10:3 Z/E mixture)

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.35 {7.35} (m, 2H; *m*-CH Ph), 7.29 {7.29} (m, 1H; *p*-CH Ph), 7.12 {7.12}

(m, 2H; o-CH Ph), 6.60 {6.16} (d, ${}^{4}J(H,H) = 1.4 \text{ Hz}$ {1.4 Hz}, 1H; 2-H), 4.10 {4.02} (m, 1H; 4-H), 3.61 {3.80} $(t \{m\}, {}^{3}J(H,H) = 8.4 \text{ Hz}, 2H; \text{ NCH}_{2}), 3.25 \{3.37\} (dd,$ ${}^{2}J(H,H) = 17.9 \text{ Hz} \{17.9 \text{ Hz}\}, {}^{3}J(H,H) = 7.7 \text{ Hz} \{7.7 \text{ Hz}\},$ 1H; cis-5-H₂), 2.93 {2.85} (q, ${}^{3}J(H,H) = 7.4 \text{ Hz} \{7.4 \text{ Hz}\},$ 2H; SCH₂CH₃), 2.67 {2.87} (dd, ${}^{2}J(H,H) = 17.9$ Hz $\{17.9 \text{ Hz}\}, \ {}^{3}J(\text{H},\text{H}) = 2.6 \text{ Hz} \ \{2.6 \text{ Hz}\}, \ 1\text{H}; \ trans-5-\text{H}_{2}\},$ 1.68 {1.76} (m, 2H; NCH₂CH₂CH₃), 1.36 {1.31} (t, ${}^{3}J(H,H) = 7.4 \text{ Hz} \{7.4 \text{ Hz}\}, 3H; \text{SCH}_{2}CH_{3}), 0.94 \{1.02\} (t,$ ${}^{3}J(H,H) = 7.4 \text{ Hz} \{7.4 \text{ Hz}\}, 3H; \text{ NCH}_{2}CH_{2}CH_{3}; {}^{13}C$ NMR (100 MHz, CDCl₃, 300 K): δ 202.0 and 199.1 {202.0 and 198.8} [C_q, 1:4, trans- and cis-CO; W(CO)₅], 183.7 {185.5} (C_q; C=N), 172.9 {174.8} (C_q; C3), 140.8 {140.6} (C_a; *i*-C Ph), 129.1 {129.0} (CH; *m*-C Ph), 127.8 {127.7} (CH; p-C Ph), 127.1 {127.1} (CH; o-C Ph), 126.8 {115.5} (CH; C2), 67.4 {67.9} (CH₂; NCH₂), 52.5 {51.2} (CH; C4), 40.0 {49.5} (CH₂; C5), 27.1 {26.8} (SCH₂CH₃), 22.8 {23.3} (NCH₂CH₂CH₃), 13.2 {13.0} (SCH₂CH₃), 11.1 {11.1} (NCH₂CH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2065.8$ (20), 1963.5 (5), 1923.1 (100), 1909.3 (70) $[v(C \equiv O)]$, 1595.5 (5), 1545.2 (5) [v(C = C) and v(C = N)]; IR (diffuse reflexion) $[\text{cm}^{-1} (\%)]$: $\tilde{v} = 2064.8$ (10), 1967.4 (3), 1865.0 (100), 1591.3 (20), 1535.7 (30), 1493.5 (7), 1474.8 (5), 1453.9 (10), 1319.9 (8), 1073.3 (4); HRMS (ESI) calcd for $C_{21}H_{21}NSO_5WNa [M + Na]^+$: 606.0543; found: 606.0534.



4.13.4. Data for 20g

¹H NMR (500 MHz, CDCl₃, 298 K): δ 10.45 (s, br, 1H; NH), 8.00 (d, ${}^{3}J(H,H) = 15.4$ Hz, 1H; PhCH), 7.52 (m, 2H; o-CH Ph), 7.34 (m, 1H; p-CH Ph), 7.28 (m, 2H; m-CH Ph), $7.02 \text{ (m, }^{3}J(\text{H,H}) = 15.4 \text{ Hz}, 1\text{H}; \text{PhCH}=CH), 3.79 \text{ (m, 2H};$ NCH₂), 1.81 (m, 2H; NCH₂CH₂), 1.03 (t, 3H; NCH₂CH₂CH₃); ¹³C NMR (125 MHz, CDCl₃, 298 K): δ 203.3 and 197.4 [each C_q , 1:4, *trans*- and *cis*-CO; W(CO)₅], 190.8 (C_q , br; W=C); [compare: 191.0; W(CO)₆], 153.3 (C_q; C4), 149.9 (CH; C6), 134.2 (C_q; *i*-C Ph), 131.1 (CH; p-CH Ph), 129.0 (CH; m-CH Ph), 128.7 (CH; o-CH Ph), 121.7 (CH; C5), 111.4 (C_a; C3), 49.5 (NCH₂), 22.2 (NCH₂CH₂), 11.4 (NCH₂CH₂CH₃); IR (dichloromethane) $[\text{cm}^{-1} (\%)]: \tilde{v} = 3377.3 \text{ (5)} [v(\text{N-H})],$ 2083.0 (2), 1993.2 (25), 1929.5 (100) [v(C=O)], 1625.6 (10), 1550.1 (10), 1470.4 (10), 1450.1 (10); IR (diffuse reflexion) $[\text{cm}^{-1}(\%)]$: $\tilde{v} = 3358.1 (10) [v(\text{N-H})], 2083.5 (1), 2002.5$ (40), 1968.8 (20), 1899.0 (100), 1870.7 (75), 1845.7 (40) [v(C=O)], 1622.9 (10), 1555.3 (30); HRMS (ESI) calcd for $C_{19}H_{15}NO_5WNa$ $[M + Na]^+$: 544.0355; found: 544.0346.

4.14. Pentacarbonyl[allyl-(3-ethylsulfanyl-4-phenylcyclopent-2-enylidene)- amine, N-W]tungsten(0) [(1Z)-19h and (1E)-19h], pentacarbonyl[3-(allylamino)-5phenyl-penta-1,2,4-triene-1- ylidene]tungsten(0) (20h)

(E)-N-Allyl-3-phenyl-acrylimidoyl chloride (5h), generated from (2E)-N-allyl-3-phenyl acrylamide (4h, 372 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) was reacted with pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (1b, 412 mg, 1.0 mmol). Reaction with triethylamine (808 mg, 8.0 mmol) at -78 °C and work-up by flash chromatography at 20 °C on silica gel $(40 \times 1 \text{ cm}, 1:1 \text{ n-pentane/diethyl ether})$ afforded a red compound ($R_{\rm f} = 0.2$ in 10:1 *n*-pentane/diethyl ether) which was thermolabil and was transformed into compound (Z)-19h (128 mg, 22%, $R_f = 0.6$ in 10:1 *n*-pentane/diethyl ether, yellow oil) within short time. Compounds (Z)-19h and (E)-19h $(R_{\rm f} = 0.5 \text{ in } 10:1 \text{ n-pentane/diethyl ether, yellow oil) were in}$ equilibrium in 7:10 E/Z-ratio after 14 d, 20 °C in CDCl₃. A more polar dark red to violet fraction yielded compound **20h** (55 mg, 11%, $R_f = 0.5$ in 1:1:1 *n*-pentane/diethyl ether/dichloromethane, violet oil).



4.14.1. Data for(Z)-19h{(E)-19h} (obtained from a 10:4 mixture)

¹H NMR (500 MHz, CDCl₃, 298 K) δ 7.35 {7.35} (m, 2H; m-CH Ph), 7.30 {7.30} (m, 1H; p-CH Ph), 7.11 {7.13} (m, 2H; *o*-CH Ph), 6.66 {6.11} (d, ${}^{4}J(H,H) = 1.4$ Hz {1.4 Hz}, 1H; 2-H), 5.83 {5.89} (ddt, ${}^{3}J(H,H) = 17.3$ Hz $\{17.3 \text{ Hz}\}, {}^{3}J(\text{H},\text{H}) = 10.5 \text{ Hz} \{10.5 \text{ Hz}\}, {}^{3}J(\text{H},\text{H}) = 4.7 \text{ Hz}$ $\{4.7 \text{ Hz}\}, 1\text{H}; C\text{H}_2CH=C\text{H}_2), 5.28 \{5.34\} (ddt, {}^2J(\text{H},\text{H}) =$ 1.1 Hz {1.1 Hz}, ${}^{3}J(H,H) = 10.5$ Hz {10.5 Hz}, ${}^{4}J(H,H) =$ 1.9 Hz {1.9 Hz}, 1H; cis-CH₂CH=CH₂), 5.11 {5.14} (ddt, ${}^{2}J(H,H) = 1.1 \text{ Hz} \{1.1 \text{ Hz}\}, {}^{3}J(H,H) = 17.3 \text{ Hz} \{17.3 \text{ Hz}\},$ ${}^{4}J(H,H) = 1.9 \text{ Hz} \{1.9 \text{ Hz}\}, 1H; trans- CH_2CH=CH_2),$ 4.38 {4.55} (m, 2H; NCH₂), 4.10 {4.06} (m, 1H; 4-H), 3.25 {3.46} (dd, ${}^{2}J(H,H) = 18.2 \text{ Hz} \{18.2 \text{ Hz}\}, {}^{3}J(H,H) =$ 7.6 Hz $\{7.6 \text{ Hz}\}, 1\text{H}; cis-5-\text{H}_2\}, 2.96 \{2.82\}$ (q, 2H; SCH_2CH_3 , 2.67 {2.96} (dd, ²J(H,H) = 18.2 Hz {18.2 Hz}, ${}^{3}J(H,H) = 2.6 \text{ Hz} \{2.6 \text{ Hz}\}, 1\text{H}; trans-5-H_{2}\}, 1.38 \{1.30\}$ $(t, {}^{3}J(H,H) = 7.4 \text{ Hz} \{7.4 \text{ Hz}\}, 3H; \text{ SCH}_{2}CH_{3}); {}^{13}C \text{ NMR}$ (100 MHz, CDCl₃, 300 K): δ 202.2 and 199.0 {202.1 and 198.7} [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 185.6 {185.2} (C_q ; C=N), 174.2 {175.7} (C_q ; C3), 140.5 {140.4} (C_q; *i*-C Ph), 132.3 {133.0} (CH; CH₂CH=CH₂), 129.1 {129.1} (CH; m-C Ph), 127.8 {127.8} (CH; p-C Ph), 127.2 {127.2} (CH; o-C Ph), 126.3 {116.2} (CH; C2), 116.6 {116.9} (CH₂; CH₂CH=CH₂), 67.3 {67.8} (CH₂; NCH₂), 52.6 {51.3} (CH; C4), 40.2 {49.2} (CH₂; C5), 27.3 {26.9}

(SCH₂CH₃), 13.2 {12.9} (SCH₂CH₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{\nu} = 2066.4$ (25), 1966.1 (5), 1925.1 (100), 1909.3 (50) [ν (C=O)], 1594.7 (10), 1543.8 (10) [ν (C=C) and ν (C=N)]; IR (diffuse reflexion) [cm⁻¹ (%)]: $\tilde{\nu} = 2065.2$ (10), 1968.4 (4), 1881.5 (100), 1591.5 (13), 1537.8 (15), 1493.9 (3), 1454.4 (4); HRMS (ESI) calcd for C₂₁H₁₉NSO₅WNa [M + Na]⁺: 604.0386; found: 604.0379.



4.14.2. Data for 20h

¹H NMR (400 MHz, CDCl₃, 300 K): δ 10.30 (s, br; 1H; NH), 7.99 (d, ${}^{3}J(H,H) = 15.4$ Hz, 1H; PhCH), 7.53 (m, 2H; o-CH Ph), 7.34 (m, 1H; p-CH Ph), 7.29 (m, 2H; m-CH Ph), 7.04 (m, ${}^{3}J(H,H) = 15.4$ Hz, 1H; PhCH=CH), 5.90 (m, 1H; NCH₂CH), 5.38 (d, br, ${}^{3}J(H,H) = 15.4$ Hz, 1H; trans-NCH₂CHCH₂), 5.28 (d, br, ${}^{3}J(H,H) = 10.2$ Hz, 1H; cis-NCH₂CHCH₂), 4.40 (m, 2H; NCH₂).; ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 203.3 and 197.4 [each C_q, 1:4, trans- and cis-CO; W(CO)₅], (C_q; W=C), [25] 152.8 (C_q; C4), 150.3 (CH; C6), 134.3 (C_q; *i*-C Ph), 131.2 (CH; p-CH Ph), 130.7 (NCH₂CH), 129.0 (CH; m-CH Ph), 128.7 (CH; o-CH Ph), 121.8 (CH; C5), 119.8 (NCH₂CH*C*H₂), (C_q; C3), [25] 50.0 (NCH₂); IR (dichloromethane) $[\text{cm}^{-1} (\%)]: \tilde{v} = 2083.0 (1), 1993.5 (30), 1929.0$ (100) $[v(C \equiv O)]$; HRMS (ESI) calcd for C₁₉H₁₃NO₅WNa $[M + Na]^+$: 542.0198; found: 542.0181.

4.15. Pentacarbonyl[(3-ethylsulfanyl-4-phenyl-cyclopent-2enylidene)-phenyl-amine, N-W]tungsten(0) [(1 Z)-19i and (1E)-19i], pentacarbonyl[3-(phenylamino)-5-phenyl-penta-1,2,4-triene-1-ylidene]tungsten(0) (20i), (E)-3,N-diphenylthioacrylimidic acid ethyl ester (21i)

(E)-N-Phenyl-3-phenyl-acrylimidoyl chloride (5i), generated in situ from (2E)-N-phenyl-3-phenyl acrylamide (4i, 446 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) at 20 °C, 5 d, was reacted with pentacarbonyl[1-(ethylsulfanyl)eth-1-ylidene]tungsten(0) (1b, 412 mg, 1.0 mmol) and triethylamine (808 mg, 8.0 mmol) at -78 °C. Work-up at 20 °C with silica gel $(40 \times 1 \text{ cm}, 1:1 \text{ n-pentane})$ diethyl ether) gave colorless compound 21i (82 mg, 37%, $R_{\rm f} = 0.8$ in 10:1 *n*-pentane/diethyl ether) and a yellow compound (Z)-19i (13 mg, 2%, $R_f = 0.5$ in 10:1 *n*-pentane/ diethyl ether, yellow oil). Equilibration of (Z)-19i and (E)-19i ($R_{\rm f} = 0.4$ in 10:1 *n*-pentane/diethyl ether) is accompanied by strong decomposition. ¹H NMR data of (E)-19i were obtained from a sample freshly collected by chromatography. A polar deep blue fraction (30 mg, $R_f = 0.3$ in 1:1:1 *n*-pentane/diethyl ether/dichloromethane) was not characterized completely. A structure 20i is proposed on the basis of an exact mass determination.



4.15.1. Data for (Z)-19i $\{(E)$ -19i $\}$ (¹H NMR data could be collected only partially from a 20:1 mixture due to facile decomposition)

¹H NMR (500 MHz, CDCl₃, 298 K): δ 7.36 (m, 2H; *m*-CH N-Ph), 7.33 (m, 2H; m-CH Ph), 7.27 (m, 1H; p-CH Ph), 7.13 (m, 1H; p-CH N-Ph), 7.08 (m, 2H; o-CH Ph), 6.88 (m, 2H; o-CH Ph), 6.77 {5.47} (d, ${}^{4}J(H,H) = 1.4 \text{ Hz} \{1.4 \text{ Hz}\},$ 1H; 2-H), 4.03 {4.16} (m, 1H; 4-H), 2.84 {3.59} (dd, ${}^{2}J(H,H) = 18.9 \text{ Hz} \{18.3 \text{ Hz}\}, {}^{3}J(H,H) = 7.4 \text{ Hz} \{7.5 \text{ Hz}\},$ 1H; *cis*-5-H₂), 3.02 {2.58} (q, ${}^{3}J(H,H) = 7.4$ Hz {7.6 Hz}, 2H; SCH₂CH₃), 2.28 {3.06} (dd, ²J(H,H) = 18.9 Hz {18.3 Hz}, ${}^{3}J(H,H) = 2.8$ Hz {2.6 Hz}, 1H; trans-5-H₂), 1.42 {1.13} (t, ${}^{3}J(H,H) = 7.4 \text{ Hz} \{7.6 \text{ Hz}\}, 3H; \text{ SCH}_{2}CH_{3}$); ¹³C NMR (125 MHz, CDCl₃, 298 K): δ 203.2 and 198.9 $[C_q, 1:4, trans- and cis-CO; W(CO)_5], 185.5 (C_q; C=N),$ 177.0 (C_q; C3), 155.7 (C_q; *i*-C N-Ph), 140.3 (C_q; *i*-C CH-Ph), 129.5 and 129.4 (CH; m-C N-Ph), 129.1 (CH; m-C CH-Ph), 127.8 (CH; p-C CH-Ph), 127.1 (CH; o-C CH-Ph), 125.9 (CH; p-C N-Ph), 125.0 (CH; C2), 120.6 and 120.5 (each CH; o-C N-Ph), 52.7 (CH; C4), 42.1 (CH₂; C5), 27.4 (SCH₂CH₃), 13;2 (SCH₂CH₃); IR (cyclohexane) $[\text{cm}^{-1} (\%)]: \tilde{v} = 2067.0 (25), 1970.6 (10), 1941.1 (50),$ 1929.1 (100), 1923.4 (80), 1907.7 (50) [v(C=O)], 1585.5 (10), 1538.7 (10) [v(C=C) and v(C=N)]; HRMS (ESI) calcd for $C_{24}H_{19}NSO_5WNa$ $[M + Na]^+$: 640.0378; found: 640.0378.



4.15.2. Data for 20i

HRMS (ESI) calcd for $C_{22}H_{12}NO_5W [M - H]^-$: 554.0223; found: 554.0242.



4.15.3. Data for 21i

¹H NMR (400 MHz, CDCl₃, 300 K): δ 7.43 (d, ³*J*(H,H) = 16.3 Hz, 1H; PhC*H*), 7.30 (m, 7 H; *m*-CH N-Ph and *o*-, *m*-, *p*-CH Ph), 7.08 (m, 1H; *p*-CH N-Ph), 6.83 (m, 2H; *o*-CH N-Ph), 6.69 (d, ³*J*(H,H) = 16.3 Hz, 1H; PhCHC*H*), 3.12 (q, ³*J*(H,H) = 7.4 Hz, 2H; SCH₂), 1.38 (t, ³*J*(H,H) = 7.4 Hz; SCH₂CH₃); ¹³C NMR (100 MHz, CDCl₃, 300 K): δ 163.4 (C_q; C=N), 150.5 (C_q; *i*-C N-Ph), 137.8 (CH; Ph-CH), 135.4 (C_q; *i*-C Ph), 129.3, 128.8, 128.7, 127.5 (each CH, 1:2:2:2; *m*-C N-Ph and *o*-, *m*-, *p*-C Ph), 123.5 (CH; p-C N-Ph), 121.4 (CH; PhCH*C*H), 120.9 (CH; *o*-C N-Ph), 24.1 (SCH₂), 14.2 (SCH₂*C*H₃); IR (cyclohexane) [cm⁻¹ (%)]: $\tilde{\nu} = 1627.3$ (30), 1580.5 (100) [ν (C=N)]; HRMS (ESI) calcd for C₁₇H₁₈NS [*M* + H]⁺: 268.1154; found: 268.1144.

4.16. Pentacarbonyl[(3-ethylsulfanyl-4-methyl-cyclopent-2enylidene)-methyl- amine, N-W]tungsten(0) [(1Z)-19j and (1E)-19j]

(*E*)-*N*-Methyl-but-2-ene-1-carboximidoyl chloride (**5**), generated *in situ* from (*E*)-but-2-enonic acid methylamide (**4**j, 198 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol), was reacted with pentacarbonyl[1-(eth-ylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 412 mg, 1.0 mmol) and triethylamine (808 mg, 8.0 mmol) at -78 °C. Work-up at 20 °C on silica gel (40 × 1 cm, 1:1 *n*-pentane/diethyl ether) afforded a red thermolabile compound ($R_f = 0.3$ in 1:1 *n*-pentane/diethyl ether) which was transformed into yellow compound (*Z*)-**19**j (130 mg, 26%, $R_f = 0.7$ in 10:1 *n*-pentane/diethyl ether, yellow oil) after a short time. A 2:3 equilibrium (*Z*)-**19**j and (*E*)-**19**j was achieved at 20 °C, 14 d in CDCl₃ ($R_f = 0.6$ in 10:1 *n*-pentane/diethyl ether).



4.16.1. Data for (Z)-19j $\{(E)$ -19j $\}$ (obtained from a 10:4 mixture)

¹H NMR (500 MHz, CDCl₃, 298 K) δ 6.35 {6.07} (d, ${}^{4}J(H,H) = 1.3 \text{ Hz} \{1.3 \text{ Hz}\}, 1H; 2-H, \text{ NOE} (+) \text{ with SC}H_{2}$ {NOE (+) with NCH₃ and SCH₂}), 3.52 {3.58} (s, 3H; NCH₃, NOE (+) with 5-H₂ {NOE (+) with 2-H}), 3.06 $\{3.04\}$ ("t", ³*J*(H,H) = 7.1 Hz, 1H; *cis*-5-H₂), 2.95 $\{3.11\}$ (m, 1H; CHCH₃), 2.97 {2.92} (q, 2H; SCH₂CH₃), 2.25 $\{2.45\}$ ("d", ²J(H,H) = 17.7 Hz $\{17.7 \text{ Hz}\}$, 1H; trans-5-H₂, NOE (+) with NCH₃, cis-5-H₂ and 4-H NOE (+) with 4-H and *cis*-5-H₂), 1.43 {1.40} (t, ${}^{3}J(H,H) = 7.4$ Hz, 3H; SCH₂CH₃), 1.24 {1.26} (d, ${}^{3}J(H,H) = 7.1$ Hz, 3 H; CHCH₃); ¹³C NMR (125 MHz, CDCl₃, 298 K): δ 202.6 and 199.0 {202.5 and 198.9} [each C_q, 1:4, trans- and cis-CO; W(CO)₅], 184.5 {184.0} (C_a; C=N), 175.2 {177.3} (C_a; C3), 124.7 {114.0} (CH; C2), 53.2 {53.5} (NCH₃), 41.4 {40.2} (CH; C4), 39.1 {47.8} (CH₂; C5), 26.9 {26.7} (SCH₂CH₃), 20.6 {20.5} (CHCH₃), 13.3 {13.1} (SCH₂CH₃); IR (cyclohexane) $[cm^{-1} (\%)]$: $\tilde{v} = 2066.3 (15), 1964.6 (5),$ 1924.7 (100), 1909.7 (40) [v(C=O)], 1606.7 (5), 1545.5 (5) [v(C=C) and v(C=N)]; HRMS (ESI) calcd for $C_{14}H_{15}NSO_5WNa [M + Na]^+$: 516.0072; found: 516.0057.

4.17. Pentacarbonyl[(3-ethylsulfanyl-4-methyl-cyclopent-2enylidene)-propyl-amine, N-W]tungsten(0) [(1Z)-19k and (1E)-19k]

(*E*)-*N*-Propyl-but-2-ene-1-carboximidoyl chloride (**5**k), generated *in situ* from (*E*)-but-2-enonic acid propylamide (**4**k, 254 mg, 2.0 mmol) and phosphorous oxychloride (306 mg, 2.0 mmol) was reacted with pentacarbonyl[1-(eth-ylsulfanyl)eth-1-ylidene]tungsten(0) (**1b**, 412 mg, 1.0 mmol) and triethylamine (808 mg, 8.0 mmol) at -78 °C. Work-up at 20 °C on silica gel (40 × 1 cm, 1:1 *n*-pentane/diethyl ether) afforded yellow compound (*Z*)-**19k** (55 mg, 11%, $R_f = 0.6$ in 10:1 *n*-pentane/diethyl ether, yellow oil), which in CDCl₃ achieved a 9:10 equilibrium of (*Z*)-**19k** and (*E*)-**19k** ($R_f = 0.5$ in 10:1 *n*-pentane/diethyl ether) after 14 d, 20 °C.



4.17.1. Data for (Z)-19k{(E)-19k}

¹H NMR (400 MHz, CDCl₃, 300 K): δ 6.37 {5.96} (d, ${}^{3}J(H,H) = 1.2 \text{ Hz} \{1.2 \text{ Hz}\}, 1\text{H}; 2\text{-H}), 3.60 \{3.71\} (m, 1\text{H};$ NCH₂), 3.03 {3.14} (m, 1H; CHCH₃), 2.98 {3.09} (dd, ${}^{2}J(H,H) = 17.2 \text{ Hz} \{17.6 \text{ Hz}\}, {}^{3}J(H,H) = 7.0 \text{ Hz} \{7.1 \text{ Hz}\}, 1H; cis-5-H_2), 2.96 \{2.91\} (q, {}^{3}J(H,H) = 7.3 \text{ Hz} \{7.4 \text{ Hz}\}, 1H; cis-5-H_2)$ 2H; SCH₂CH₃), 2.28 {2.43} (dd, ${}^{2}J(H,H) = 17.2$ Hz $\{17.6 \text{ Hz}\}, \ {}^{3}J(\text{H},\text{H}) = 2.0 \text{ Hz} \ \{2.4 \text{ Hz}\}, \ 1\text{H}; \ trans-5-\text{H}_{2}\},\$ 1.66 {1.66} (m, 2H; NCH₂CH₂CH₃); 1.42 {1.40} (t. ${}^{3}J(H,H) = 7.3 \text{ Hz} \{7.4 \text{ Hz}\}, 3H; \text{ SCH}_{2}CH_{3}), 1.24 \{1.26\}$ $(d, {}^{3}J(H,H) = 7.0 \text{ Hz}, 3H; CHCH_{3}); 0.97 \{0.97\} (t, 3H;$ NCH₂CH₂CH₃); ¹³C NMR (150 MHz, CDCl₃, 300 K): δ 202.2 and 199.1 {202.3 and 198.9} [each C_q, 1:4, *trans*and cis-CO; W(CO)₅], 183.7 {183.4} (C_a; C=N), 174.9 $\{176.7\}$ (C_a; C3), 125.4 $\{114.2\}$ (CH; C2), 67.3 $\{67.6\}$ (CH; NCH₂), 41.4 {40.0} (CH; C4), 38.2 {48.3} (CH₂; C5), 26.9 {26.6} (SCH₂CH₃), 22.8 {23.2} (CH₂; NCH₂CH₂CH₃]), 20.6 {20.5} (CH₃; CHCH₃), 13.3 {13.1} (SCH₂CH₃), 11.3 {11.2} (CH₃; NCH₂CH₂CH₃); IR (cyclohexane) $[cm^{-1}(\%)]$: $\tilde{v} = 2065.6(25), 1963.2(5), 1924.2(100),$ 1909.3 (70) [v(C=O)], 1597.4 (5), 1547.1 (5) [v(C=N) and v(C=C); HRMS (ESI) calcd for $C_{16}H_{19}NSO_5WNa$ $[M + Na]^+$: 544.0386; found: 544.3384.

5. Supplementary material

CCDC 622534 and 622535 contain the supplementary crystallographic data for **13a** and **18b**. These data can be obtained free of charge via http://www.ccdc.cam.ac.uk/conts/retrieving.html, or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (+44) 1223-336-033; or e-mail: deposit@ ccdc.cam.ac.uk.

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